

# Correlation between the conformational states of $F_1$ -ATPase as determined from its crystal structure and single-molecule rotation

Daichi Okuno<sup>a</sup>, Ryo Fujisawa<sup>a</sup>, Ryota Iino<sup>a</sup>, Yoko Hirono-Hara<sup>b</sup>, Hiromi Imamura<sup>a,c</sup>, and Hiroyuki Noji<sup>a,1</sup>

<sup>a</sup>The Institute of Scientific and Industrial Research, Osaka University, 8-1 Mihogaoka, Ibaraki, Osaka 567-0047, Japan; <sup>b</sup>Institute of Industrial Science, University of Tokyo, 4-6-1 Komaba, Tokyo 153-8505, Japan; and <sup>c</sup>Precursory Research for Embryonic Science and Technology, Japan Science and Technology Agency, 5 Sanbancho, Tokyo 102-0075, Japan

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$F_1$ -ATPase is a rotary molecular motor driven by ATP hydrolysis that rotates the  $\gamma$ -subunit against the  $\alpha_3\beta_3$  ring. The crystal structures of  $F_1$ , which provide the structural basis for the catalysis mechanism, have shown essentially 1 stable conformational state. In contrast, single-molecule studies have revealed that  $F_1$  has 2 stable conformational states: ATP-binding dwell state and catalytic dwell state. Although structural and single-molecule studies are crucial for the understanding of the molecular mechanism of  $F_1$ , it remains unclear as to which catalytic state the crystal structure represents. To address this issue, we introduced cysteine residues at  $\beta$ E391 and  $\gamma$ R84 of  $F_1$  from thermophilic *Bacillus* PS3. In the crystal structures of the mitochondrial  $F_1$ , the corresponding residues in the ADP-bound  $\beta$  ( $\beta_{DP}$ ) and  $\gamma$  were in direct contact. The  $\beta$ E190D mutation was additionally introduced into the  $\beta$  to slow ATP hydrolysis. By incorporating a single copy of the mutant  $\beta$ -subunit, the chimera  $F_1$ ,  $\alpha_3\beta_2\beta$ (E190D/E391C) $\gamma$ (R84C), was prepared. In single-molecule rotation assay, chimera  $F_1$  showed a catalytic dwell pause in every turn because of the slowed ATP hydrolysis of  $\beta$ (E190D/E391C). When the mutant  $\beta$  and  $\gamma$  were cross-linked through a disulfide bond between  $\beta$ E391C and  $\gamma$ R84C,  $F_1$  paused the rotation at the catalytic dwell angle of  $\beta$ (E190D/E391C), indicating that the crystal structure represents the catalytic dwell state and that  $\beta_{DP}$  is the catalytically active form. The former point was again confirmed in experiments where  $F_1$  rotation was inhibited by adenosine-5'-( $\beta,\gamma$ -imino)-triphosphate and/or azide, the most commonly used inhibitors for the crystallization of  $F_1$ .

ATP synthase | cross-link

The  $F_0F_1$ -ATP synthase is widely found in biological membranes such as the mitochondrial inner membrane, thylakoid membrane, and bacterial plasma membrane. It catalyzes ATP synthesis from ADP and inorganic phosphate ( $P_i$ ) at the cost of the electrochemical potential of proton or sodium ion across the membrane (1–6). The synthase is composed of a water-soluble part,  $F_1$ , and a membrane-embedded part,  $F_0$ .  $F_1$  is a rotary motor driven by ATP hydrolysis. The subunit composition of bacterial  $F_1$  is  $\alpha_3\beta_3\gamma\delta\epsilon$ . The minimum complex of the rotary motor is the  $\alpha_3\beta_3\gamma$  subcomplex that is hereinafter referred to as  $F_1^{\alpha\beta\gamma}$ . Hydrolyzing ATP,  $F_1$  rotates the  $\gamma$ -subunit against the  $\alpha_3\beta_3$  stator ring in a counterclockwise direction when viewed from the  $F_0$  (7).  $F_0$  is also a rotary motor that is driven by proton translocation across the membrane, down the electrochemical potential (8, 9). The subunit composition of bacterial  $F_0$  is  $ab_2c_{10-15}$  in which the  $c_{10-15}$  oligomer ring rotates against the  $ab_2$  stator. These 2 rotary motors are connected through the central and peripheral stalks. Under ATP synthesis conditions where the proton electrochemical potential is dominant,  $F_0$  forcibly rotates the  $\gamma$ -subunit of  $F_1$  in the reverse direction (clockwise when viewed from  $F_0$ ). This reverse rotation of  $F_1$  results in the reverse chemical reaction of ATP hydrolysis, i.e., ATP synthesis. Conversely, under ATP hydrolysis conditions,  $F_1$  rotates  $\gamma$  in the anticlockwise direction with the  $c_{10-15}$  ring, forcing  $F_0$  to pump protons against the proton electrochemical potential.

The atomic-level structure of  $F_1$  was first described in 1994, using a crystal of bovine mitochondrial  $F_1$  ( $MF_1$ ) prepared in the presence of ADP, adenosine-5'-( $\beta,\gamma$ -imino)-triphosphate (AMP-PNP), and sodium azide ( $NaN_3$ ) (10). This structure, referred to as the “reference structure,” revealed that 3  $\alpha$ -subunits and 3  $\beta$ -subunits are alternately aligned to form a hetero hexamer ring and the  $\gamma$ -subunit is set into the central cavity of the  $\alpha_3\beta_3$  ring. The catalytic sites are located at each  $\alpha\beta$  interface, mainly on the  $\beta$ -subunit. Each  $\beta$  is in a different conformational state depending on the bound substrate; one binds to AMP-PNP ( $\beta_{TP}$ ), another to ADP ( $\beta_{DP}$ ), and the third to none ( $\beta_{empty}$ ). Both  $\beta_{TP}$  and  $\beta_{DP}$  are in the closed conformation where the C-terminal domain swings toward the nucleotide-binding domain to close the cleft between these domains. As a result, these  $\beta$ -subunits wrap the bound nucleotide tightly. In contrast,  $\beta_{empty}$  adopts an open conformation to weaken the affinity to the nucleotide. These structural features agree well with the binding change mechanism (5), which assumes that each catalytic site is in a different catalytic state and the interconversion of catalytic states drives the rotary motion of the  $\gamma$ -subunit, although there are some inconsistencies (11). Since this work, many crystal structures of  $MF_1$  with different chemical inhibitors have been reported. The crystal structure that differs most from the reference structure is that which has  $ADP\cdot AlF_4^-$ , where  $\beta_{empty}$  binds to ADP to adopt the half-closed conformation and the  $\gamma$  is twisted by  $-20^\circ$  (12). However, all other  $MF_1$  structures have very similar conformations to the reference structure. Thus, the crystal structures of  $MF_1$  essentially represent a certain stable conformational state of  $F_1$  except for the  $ADP\cdot AlF_4^-$ -bound  $MF_1$  structure.

In contrast, single-molecule studies on the  $\gamma$  rotation of  $F_1$  have revealed that  $F_1$  has 2 distinct stable conformations. Since the observation that  $F_1$  performs a  $120^\circ$  step rotation of  $\gamma$  upon hydrolysis of 1 ATP, intensive attempts have been made to resolve the  $120^\circ$  step into multiple substeps to allow better understanding of how the elementary catalytic steps are coupled with the mechanical rotation. In many of these studies including this one,  $F_1^{\alpha\beta\gamma}$  from thermophilic *Bacillus* PS3 ( $TF_1$ ) have been used because  $TF_1$  is stable under the harsh conditions of the single-molecule experiments and can be genetically modified. Thus far, 2 substeps have been found: the  $80^\circ$  and  $40^\circ$  substeps. High-speed imaging of the rotation (13) and a study of a mutant  $F_1$  having a noticeably low ATP hydrolysis rate (14) revealed that the  $80^\circ$  substep is induced by ATP binding and that the  $40^\circ$  substep was initiated after hydrolysis

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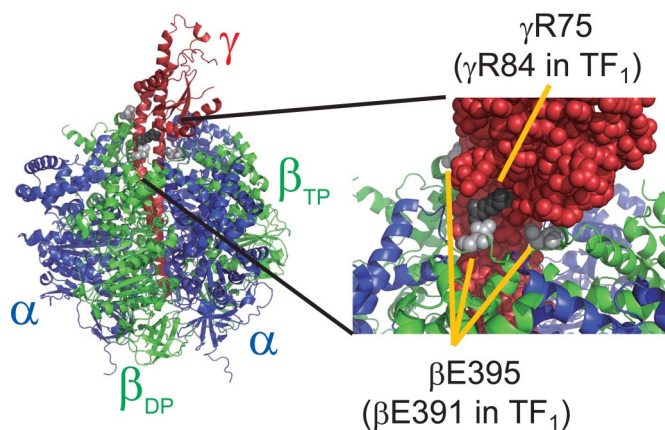
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<sup>1</sup>To whom correspondence should be addressed. E-mail: hnoji@sanken.osaka-u.ac.jp.

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**Fig. 1.** Spatial positions of  $\beta$ E395 (light gray spheres) and  $\gamma$ R75 (dark gray spheres) of  $MF_1$  in the crystal structure (Protein Data Bank ID code 1e79) (26). These residues correspond to  $\beta$ E391 and  $\gamma$ R84 of  $TF_1$ . The protruding part of the  $\gamma$  is the  $F_0$  binding side. The  $\alpha$ -subunit is shown in blue, the  $\beta$ -subunit is green, and the  $\gamma$ -subunit is red. To show the  $\beta$ E395 and  $\gamma$ R75 residues, R33, D74, D110, R113, R133, R134, and P135 residues of the  $\gamma$  were removed. The figure was produced with Pymol.

of bound ATP. Hereinafter, the 2 conformational states before the  $80^\circ$  or  $40^\circ$  substeps are referred to as the “binding dwell state” and “catalytic dwell state,” respectively. Recent studies have suggested that ADP release and  $P_i$  release occur at the binding dwell angle and catalytic dwell angle, respectively (15, 16). More recently, a temperature-sensitive reaction was found at the binding dwell angle in a rotation assay at low temperature (16).

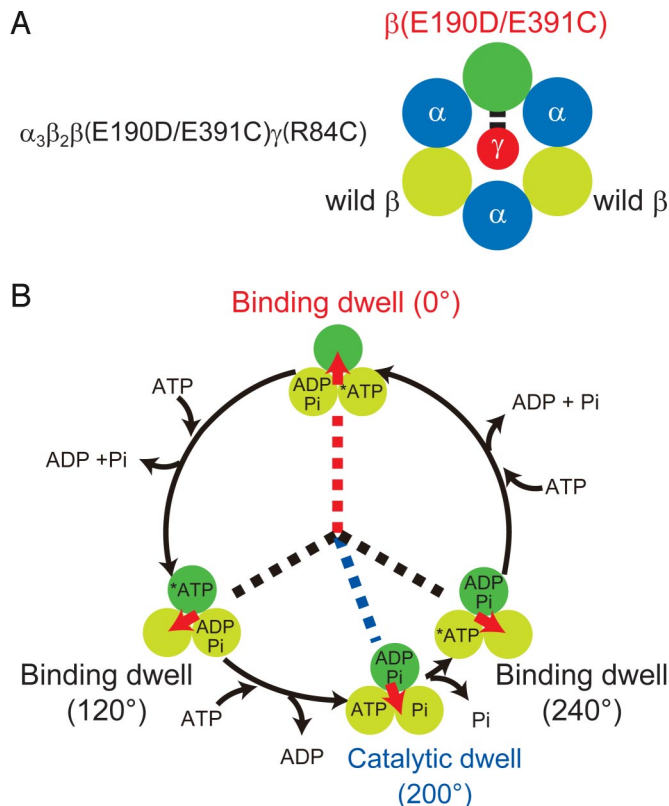
Thus, single-molecule studies have revealed that  $F_1$  has 2 stable conformational states, whereas  $F_1$  adopts essentially 1 specific stable conformation in the crystal structure. Which conformational state do the crystal structures of  $F_1$  represent, binding dwell state, catalytic dwell state, or another new state? On this issue, there are some interesting results. The analysis of fluorescence resonance energy transfer (FRET) between fluorescent probes attached to  $\beta$  and  $\gamma$  suggests that the crystal structure probably corresponds to the catalytic dwell state (17). However, FRET measurement provides the relative distance between the fluorophores, but not the precise position in the 3D structure. Therefore, this result is not conclusive. The reference structure was recently revised to show that  $\beta_{DP}$  binds to  $N_3^-$  at the  $\gamma$ -phosphate binding position (18), implying that  $F_1$  in the crystal structure is in the azide-stabilized form, while the feature of this structure was later revealed to be almost consistent with that of the catalytic ground state  $MF_1$  (19). Taking into account that  $F_1$  pauses at the catalytic angle when in the ADP-inhibited form (20), it seems plausible that the crystal structure is in the catalytic dwell state. However, there are no experimental results that clarify this

**Table 1. Distance between  $\gamma$ -carbons of  $\beta$ E395 and  $\gamma$ R75 in crystal structures of  $MF_1$**

Protein Data Bank ID code	Ref.	Distance, Å		
		$\beta_{TP}$	$\beta_{DP}$	$\beta_{empty}$
1e79	26	11.2	6.1	25.1
1h8e	12	11.1	6.0	17.8
1w0j	22	11.2	6.0	24.8
2ck3	18	10.8	5.7	ND
2jdi	19	11.0	5.8	ND
		$11.0 \pm 0.2^*$	$5.9 \pm 0.2^*$	$22.6 \pm 4.1^*$

ND, not determined because E395 in  $\beta_{empty}$  was not visible in these structures.

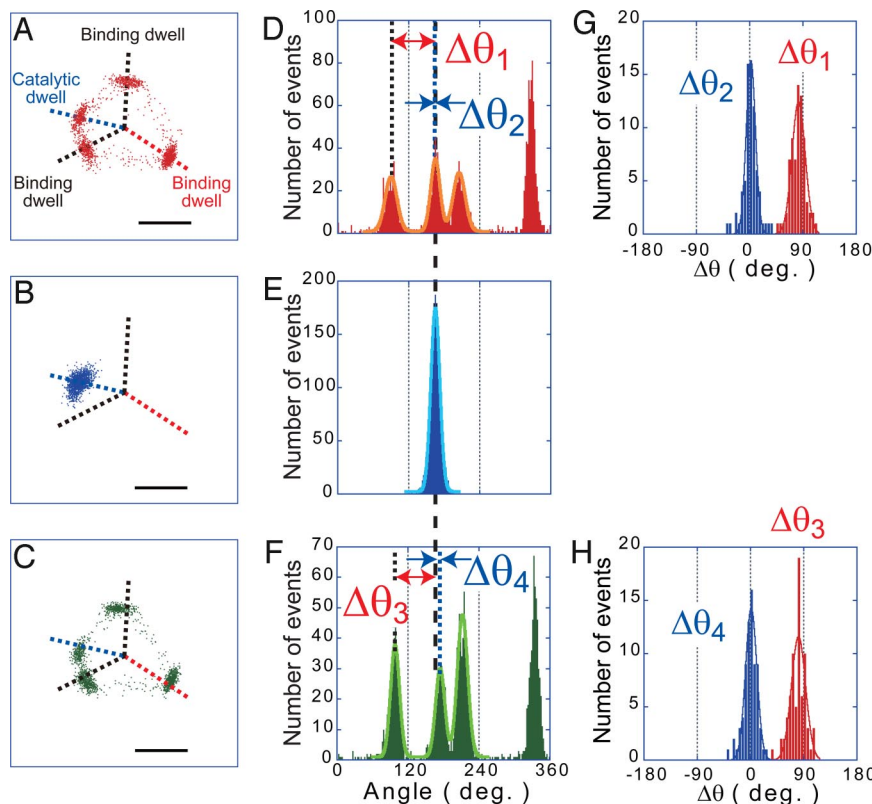
\*Average  $\pm$  SD.



**Fig. 2.** The chimera  $F_1^{\alpha\beta\gamma}$ ,  $\alpha_3\beta_2\beta(E190D/E391C)\gamma(R84C)$ . (A) Schematic image of the chimera  $F_1^{\alpha\beta\gamma}$ ,  $\alpha_3\beta_2\beta(E190D/E391C)\gamma(R84C)$  used in the cross-link experiment. Short black prongs denote cysteine residues at  $\beta$ E391 and  $\gamma$ R84 for disulfide bond formation. (B) Circular diagram of the reaction and rotation of the chimera  $F_1^{\alpha\beta\gamma}$ . At  $0^\circ$ , the mutated  $\beta$ ,  $\beta(E190D/E391C)$ , binds ATP (red dotted line) and makes the long catalytic dwell at  $200^\circ$  (blue dotted line) caused by slowed ATP hydrolysis by the  $\beta$ E190D mutation (29). The wild-type  $\beta$ -subunits bind ATP at  $120^\circ$  or  $240^\circ$  (black dotted lines). The catalytic dwells by the wild-type  $\beta$ -subunits at  $80^\circ$  or  $320^\circ$  were too short (only 1 ms) to detect with the present recording system (33 ms per frame) and are not shown.

point. In fact, a common view on the nature of crystal structure of the  $F_1$  has not been established yet: some postulate it to be in the binding dwell state (21, 22) and others think it is the catalytic dwell state (23, 24).

With an aim to addressing this issue, we generated a mutant  $F_1^{\alpha\beta\gamma}$  from *Bacillus* PS3 in which cysteine residues were introduced at  $\beta$ E391 and  $\gamma$ R84, respectively (25). In the crystal structures of bovine  $MF_1$ , corresponding residues in the ADP-bound  $\beta$  ( $\beta_{DP}$ ) and the  $\gamma$  are in direct contact (18, 19, 22, 26). The residue of  $\gamma$ R84 is a part of the “ionic track” (25, 27), which is the distinctive zonation of positively-charged residues around the axis of  $\gamma$ . It is postulated that  $\beta$  bends and unbends its conformation, tracing the ionic track with the negative charges of  $\beta$ D394 and  $\beta$ E395 (for  $MF_1$ ) so as to convert the bending motion of  $\beta$  into the rotary motion of  $\gamma$  (27). The importance of the ionic track is supported by the analysis of mutant  $F_1^{\alpha\beta\gamma}$  in which  $\beta$ E391 and  $\gamma$ R84 are substituted with cysteine (25). Thus, the direct contact between  $\beta$ E391 and  $\gamma$ R84 is the representative  $\beta_{DP}$ - $\gamma$  interaction in the conformational state of the crystal structure. In this study,  $F_1^{\alpha\beta\gamma}$  with  $\beta$ E391C and  $\gamma$ R84C mutations was investigated in a single-molecule rotation assay to observe where it pauses when the introduced cysteine residues form a disulfide bond, thereby fixing the motor in the conformational state corresponding to the conformational state of the crystal structures. In addition to the cross-link experiment, the correlation between the crystal structures and the single-molecule experiment was studied by investigating at which angle  $F_1^{\alpha\beta\gamma}$  stops rotation in the



**Fig. 3.**  $\beta$ - $\gamma$  cross-linking experiment in the single-molecule rotation assay. (A–C) The centroid traces of the rotation of a  $\alpha_3\beta_2\beta(\text{E190D}/\text{E391C})\gamma(\text{R84C})$  molecule. Data are derived from 60-s recordings. (A) The rotation at 200 nM ATP with 3 binding dwells by all  $\beta$ -subunits and 1 catalytic dwell caused by the  $\beta(\text{E190D}/\text{E391C})$ . (B) Pause of rotation after cross-linking by disulfide bond between  $\beta\text{E391C}$  and  $\gamma\text{R84C}$ . (C) Rotation resumed after the reduction of the disulfide bond. (Scale bars: 100 nm.) (D–F) Histograms of rotary angle of traces shown in A–C. The angular positions of the pauses were determined by fitting the data with Gaussian curves (orange, light blue, and yellow green lines in D, E, and F, respectively). Then, the angular distance of the cross-link pause (dashed line in E) from the binding dwell angle on the clockwise side ( $\Delta\theta_1$ ) or the catalytic dwell angle of  $\beta\text{E391C}$  ( $\Delta\theta_2$ ) of the rotation before cross-link were determined. These distances were again determined by comparison with the rotation after reduction as  $\Delta\theta_3$  and  $\Delta\theta_4$ . (G and H) Histograms of the angular distances ( $\Delta\theta_1$  to  $\Delta\theta_4$ ).  $n = 72$  (36 molecules). The means  $\pm$  SD for  $\Delta\theta_1$ ,  $\Delta\theta_2$ ,  $\Delta\theta_3$ , and  $\Delta\theta_4$  were determined by Gaussian curve fitting (red and blue lines) to be  $82.7 \pm 15.9^\circ$ ,  $2.0 \pm 11.0^\circ$ ,  $81.3 \pm 17.0^\circ$ , and  $1.4 \pm 13.5^\circ$ , respectively.

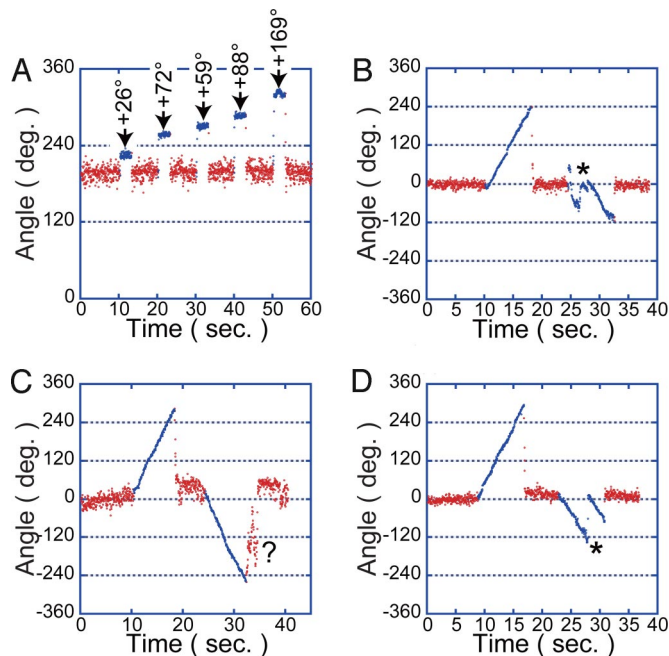
presence of AMP-PNP and/or  $\text{NaN}_3$ , the most commonly used chemicals for the crystallization of  $\text{F}_1$ .

## Results

**A Chimera  $\text{F}_1^{\alpha\beta\gamma}$ ,  $\alpha_3\beta_2\beta(\text{E190D}/\text{E391C})\gamma(\text{R84C})$ .** A mutant  $\text{F}_1^{\alpha\beta\gamma}$  from *Bacillus* PS3 was prepared in which cysteine residues were introduced at  $\beta\text{E391}$  and  $\gamma\text{R84}$ , which correspond to the  $\beta\text{E395}$  and  $\gamma\text{R75}$  of bovine  $\text{MF}_1$ , respectively. In the crystal structure of bovine  $\text{MF}_1$ , these 2 residues are in direct contact (Fig. 1). Although the whole structure of  $\text{TF}_1$  has not been solved, it is most likely that  $\text{TF}_1$  structure is very similar to  $\text{MF}_1$ , and that  $\beta\text{E391}$  and  $\gamma\text{R84}$  of  $\text{TF}_1$  are also in a very close proximity when  $\text{TF}_1$  takes the conformational state corresponding to the crystal structure of  $\text{MF}_1$ . This is because  $\text{F}_1$  from other species shows structural features essentially identical to that of  $\text{MF}_1$  even if the amino acid sequence is not highly homologous to that of  $\text{MF}_1$ , as seen in the  $\gamma$  subunit of  $\text{F}_1$  from *Escherichia coli* ( $\text{EF}_1$ ): The structure of the  $\gamma$  of  $\text{EF}_1$  is very similar to that of  $\text{MF}_1$  (28), although the sequence homology against the  $\gamma$  of  $\text{MF}_1$  is low (only  $\approx 30\%$ ) the same as the  $\gamma$  of  $\text{TF}_1$ . Furthermore, the sequences around  $\beta\text{E391}$  and  $\gamma\text{R84}$  are highly conserved in many species, suggesting the high structural conservation of these regions (Figs. S1 and S2). When  $\beta\text{E391}$  and  $\gamma\text{R84}$  of  $\text{F}_1^{\alpha\beta\gamma}$  from *Bacillus* PS3 were replaced with cysteine residues, they efficiently ( $\approx 90\%$ ) formed a disulfide bond under oxidizing conditions within only a few minutes (25). Thus, it is reasonable to assume that  $\text{TF}_1$  has the essentially identical structural features to  $\text{MF}_1$ . In the  $\text{MF}_1$  crystal structure, the  $\beta$  that has direct contact with  $\gamma\text{R75}$  adopts the  $\beta_{\text{DP}}$  form. The  $\gamma$ -carbons of these residues that are in equivalent

positions to the sulfur atoms of cysteine residue are only  $5.9 \text{ \AA}$  away (Table 1), which is sufficiently close for disulfide bond formation. In contrast, the distances in the  $\beta_{\text{TP}}$ - or  $\beta_{\text{empty}}\text{-}\gamma$  pairs are 11 and  $23 \text{ \AA}$ , respectively. Thus, the disulfide bond would be formed in the  $\beta_{\text{DP}}\text{-}\gamma$  pair. In the single-molecule rotation assay,  $\gamma$  was actually cross-linked to  $\beta$  only when the  $\beta$  was in a specific catalytic state, revealed to be the catalytic dwell state (see below). Furthermore, E190D mutation was introduced to the  $\beta$ -subunit in addition to E391C mutation. The mutant  $\beta(\text{E190D}/\text{E391C})^*$  was reconstituted with the wild-type  $\beta$  to build a chimera  $\text{F}_1^{\alpha\beta\gamma}$ ,  $\alpha_3\beta_2\beta(\text{E190D}/\text{E391C})\gamma(\text{R84C})$ , which has a single copy of  $\beta(\text{E190D}/\text{E391C})$  (Fig. 2A). Here, the  $\beta\text{E190D}$  mutation was used as an angular position marker to determine the pause angle of cross-linked  $\text{F}_1^{\alpha\beta\gamma}$ . Shimabukuro *et al.* (14) showed that the  $\beta\text{E190D}$  mutation severely slows the hydrolysis step and causes the long pause ( $\tau = 320 \text{ ms}$ ) at the catalytic angle. When a single copy of  $\beta$  with E190D mutation is incorporated in  $\text{F}_1^{\alpha\beta\gamma}$ , the reconstituted chimera  $\text{F}_1^{\alpha\beta\gamma}$  exhibits a transient pause at the catalytic dwell angle of  $\beta\text{E190D}$  that is  $+200^\circ$  from the angle at which the  $\beta\text{E190D}$  subunit binds to ATP (Fig. 2B) (29). Thus, a chimera  $\text{F}_1^{\alpha\beta\gamma}$  with a single  $\beta(\text{E190D})$  enables us to determine the pause angle of the cross-linked  $\text{F}_1^{\alpha\beta\gamma}$  relative to the catalytic angle of  $\beta(\text{E190D})$ . It should be noted that a chimera  $\text{F}_1^{\alpha\beta\gamma}$  with 1  $\beta(\text{E190D})$  was reported to exhibit another short pause at  $+120^\circ$  from the catalytic dwell angle of  $\beta(\text{E190D})$  (29). In this

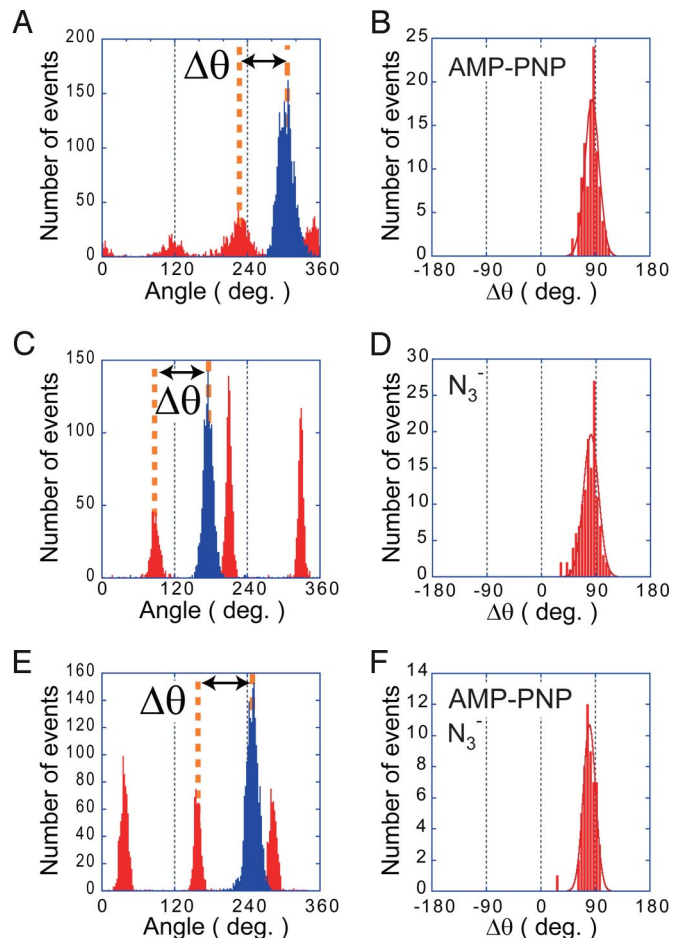
\*The designation subunit name (mutation name) indicates the subunit with the indicated mutations. The designation without parentheses indicates a mutation.



**Fig. 4.** Manipulation of cross-linked  $F_1^{\alpha\beta\gamma}$  with the magnetic tweezers. (A) Time course of the manipulation. Blue dots represent the period for the manipulation. The molecule was clamped at the indicated angle for 3 s in the forward direction and released. (B–D)  $F_1^{\alpha\beta\gamma}$  was rotated at the rate of 36° per s for near or  $>120^\circ$  in both directions. When twisted in the backward direction,  $F_1^{\alpha\beta\gamma}$  showed irregular responses. Some resisted the external magnetic field, inverted the magnetic moment of the beads instantaneously (at the point indicated by \* in B and D), and then rotated back to the inverted angle of the original magnetic field (B and D), or some paused at irregular positions (indicated by a mark in C) after being released from the field.

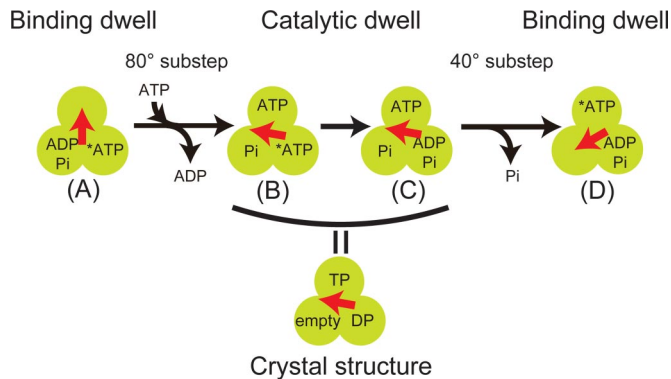
study, the short pause is neglected because the time constant of the short pause is too short ( $\tau \approx 15$  ms) for the recording rate in this study (33 ms per frame).

**Pausing Position of the Cross-Linked  $F_1^{\alpha\beta\gamma}$ .** The rotation assay of  $\alpha_3\beta_2\beta(E190D/E391C)\gamma(R84C)$  was carried out under a substrate-limiting condition, at 200 nM ATP. Because the other kinds of chimera  $F_1^{\alpha\beta\gamma}$ , which carry 2, 3, or 0 of  $\beta(E190D/E391C)$  contaminated in the rotation assay,  $\alpha_3\beta_2\beta(E190D/E391C)\gamma(R84C)$  was identified based on 2 criteria. (i) The existence of obvious pauses in every turn at the 3 binding angles,  $120^\circ$  apart from each other. (ii) The existence of the clear pause in every turn at a catalytic angle, which divides 1 of 3  $120^\circ$  steps into the  $80^\circ$  and  $40^\circ$  substeps in this order (not  $40^\circ$  and  $80^\circ$  substeps) (Fig. 3A and C) (29). In the selection based on the first criterion,  $>95\%$  of molecules were omitted from the further analysis because of ambiguous stepping. Among selected molecules, 50% showed the clear pausing at a catalytic angle and were identified as the chimera,  $\alpha_3\beta_2\beta(E190D/E391C)\gamma(R84C)$ . The remaining were mostly  $\alpha_3\beta_3\gamma(R84C)$ , of which all  $\beta$ -subunits were wild type, because the other chimera  $F_1^{\alpha\beta\gamma}$  molecules carrying 2 or 3 mutant  $\beta$ -subunits were difficult to find out because of their slow rotary motion and the reconstitution efficiency of the mutant  $\beta$  was lower than that of the wild type as reported (29). Fig. 3A–F shows a single dataset of the cross-link experiment. First, the free rotation of a  $\alpha_3\beta_2\beta(E190D/E391C)\gamma(R84C)$  molecule was observed to determine the angular position of the catalytic dwell of  $\beta(E190D/E391C)$  and the 3 binding dwell angles (Fig. 3A and D). Then, an oxidizing buffer containing 3-carboxy-4-nitrophenyl disulfide (DTNB) was injected to the flow chamber to promote a disulfide bond formation between  $\beta E391C$  and  $\gamma R84C$ . After the buffer exchange, which took several minutes, a significant fraction of the molecules ( $\approx 70\%$ ) stopped rotation



**Fig. 5.** Pausing position of  $F_1^{\alpha\beta\gamma}$  stalled by chemical inhibitors. (A) Histogram of the angular position from a single experimental dataset of the AMP-PNP inhibition experiment. After observing active rotation of a molecule at 60 nM ATP, 1  $\mu$ M AMP-PNP was infused into the reaction chamber with 60 nM ATP to stop the rotation. Red bars represent the angular position of free rotation; blue bars represent AMP-PNP inhibition. (B) Histogram of the angular deviation of the position of AMP-PNP inhibition from the binding dwell angle on the clockwise side ( $\Delta\theta$  in A). The mean was  $84.2 \pm 16.3^\circ$ .  $n = 107$  (15 molecules). (C and D) The angular position of azide ( $N_3^-$ ) inhibition. The rotation was inhibited by infusing 1 mM  $NaN_3$  with 200  $\mu$ M ATP. The mean value of  $\Delta\theta$  was  $81.9 \pm 16.3^\circ$ .  $n = 129$  (29 molecules). (E and F) The angular position when inhibited by 200  $\mu$ M AMP-PNP, 5  $\mu$ M ADP, and 1 mM  $NaN_3$ . The mean value of  $\Delta\theta$  was  $80.2 \pm 14.5^\circ$ .  $n = 54$  (9 molecules).

(Fig. 3B and E). The remaining would be molecules in which cysteine residue was modified with a biotin that blocks the disulfide bond formation. After observing a pause, a pausing  $F_1^{\alpha\beta\gamma}$  molecule was manipulated with magnetic tweezers to confirm the  $\beta$ - $\gamma$  cross-linkage (Fig. 4). Unlike ADP-inhibited  $F_1^{\alpha\beta\gamma}$ , cross-linked  $F_1^{\alpha\beta\gamma}$  never resumes rotation, even when forcibly rotated more than  $+80^\circ$ , which is sufficient to reactivate ADP-inhibited  $F_1^{\alpha\beta\gamma}$  with nearly 100% efficiency (Fig. 4A) (30). Instead, cross-linked  $F_1^{\alpha\beta\gamma}$  behaved as a twisted spring: when forcibly rotated and released, it just returned to the original pausing position. The angular velocity of return was always very fast (3.8 revolutions per s) and comparable with the ATP-driven rotation velocity at high ATP concentration ( $\approx 5$  revolutions per s). In rare cases,  $F_1^{\alpha\beta\gamma}$  exhibited irregular behaviors such as large fluctuations at an irregular position (Fig. 4B–D). Such irregular behaviors were observed in some cases where cross-linked  $F_1^{\alpha\beta\gamma}$  was rotated more than  $\pm 120^\circ$  with the magnetic tweezers. This phenomenon is probably caused by the partial unfolding of the  $\gamma$ - or  $\beta$ -subunits. When the buffer was exchanged



**Fig. 6.** The proposed reaction scheme of  $F_1$  and correlation with the crystal structure. State A represents the binding dwell state. After ATP binding and ADP release,  $F_1$  makes an  $80^\circ$  substep. Thereafter,  $F_1$  hydrolyzes the tightly bound ATP (denoted by \*) in the state transition from B to C. These states correspond to the crystal structure of  $F_1$ . After releasing  $P_i$ ,  $F_1$  makes a  $40^\circ$  substep to complete a cycle of ATP hydrolysis reaction coupled with  $120^\circ$  rotation. State D is the next binding dwell state.

with a reducing one, the disulfide bond was cleaved and  $F_1^{\alpha\beta\gamma}$  molecules resumed active rotation (Fig. 3 C and F). Molecules that did not resume rotation or significantly changed their binding dwell angles were omitted from our data analysis. The angle differences of the cross-link angle (Fig. 3E) from the nearest binding angle on the clockwise side ( $\Delta\theta_1$  in Fig. 3D) or the catalytic angle of  $\beta(E190D/E391C)$  ( $\Delta\theta_2$  in Fig. 3D) were determined. These values were also determined by the comparison with the binding and catalytic dwell angles after reduction ( $\Delta\theta_3$  and  $\Delta\theta_4$  in Fig. 3F). For this analysis, the set of the experiments was repeated a total of 72 times with 36 molecules. The histograms of  $\Delta\theta_1$  and  $\Delta\theta_2$  showed a single peak and gave mean values of  $82.7 \pm 15.9^\circ$  for  $\Delta\theta_1$  and  $2.0 \pm 11.0^\circ$  for  $\Delta\theta_2$  (Fig. 3G). The histograms of  $\Delta\theta_3$  and  $\Delta\theta_4$  also gave essentially the same values,  $81.3 \pm 17.0^\circ$  for  $\Delta\theta_3$  and  $1.4 \pm 13.5^\circ$  for  $\Delta\theta_4$ , showing the reproducibility of the experiment (Fig. 3H). These results show that the cross-linked  $F_1^{\alpha\beta\gamma}$  pauses at the catalytic dwell angle of  $\beta(E190D/E391C)$ . This means that the reference structure of  $MF_1$  represents the conformational state of  $F_1$  in the catalytic dwell state, and that  $\beta_{DP}$  conformation represents the catalytically active state that executes ATP hydrolysis. Considering that the standard deviations are much smaller than the magnitude of the  $40^\circ$  substep, cross-linked  $F_1^{\alpha\beta\gamma}$  will not be stable at the binding angle.

Other chimera  $F_1^{\alpha\beta\gamma}$  molecules,  $\alpha_3\beta(E190D)_2\beta(E391C)\gamma(R84C)$  and  $\alpha_3\beta(E190D)\beta(E391C)_2\gamma(R84C)$ , were also examined to confirm that  $\gamma R84C$  can form a disulfide bond with  $\beta E391C$ , which does not have the  $\beta E190D$  mutation. The rotation of  $\alpha_3\beta(E190D)_2\beta(E391C)\gamma(R84C)$  was rarely seen because of the low reconstitution efficiency of  $\beta(E190D)$  as described above. The cross-linked chimera paused at either of the 2 catalytic angles that did not correspond to the catalytic angle of  $\beta E190D$  (Fig. S3). Thus, it was verified that  $\beta E190D$  mutation does not affect the pause position of cross-linked  $F_1^{\alpha\beta\gamma}$ .

**Pause Angle of  $F_1^{\alpha\beta\gamma}$  Inhibited by AMP-PNP or  $N_3^-$ .** To crystallize  $F_1$ , chemical inhibitors are often used to stabilize it in a specific conformational state. Another complementary method to correlate the crystal structure of  $F_1$  and the substeps found in the single-molecule rotation assay is to analyze the pausing angle of  $F_1^{\alpha\beta\gamma}$  inhibited by these inhibitors in the rotation assay. AMP-PNP and  $N_3^-$  are the chemical inhibitors most often used for the crystallization of  $F_1$ . Therefore, AMP-PNP and  $N_3^-$  were used to stop the rotation of  $F_1^{\alpha\beta\gamma}$  in this experiment (Fig. 5). As in the cross-linking experiment, the  $120^\circ$  stepping rotation of a wild-type  $F_1^{\alpha\beta\gamma}$  molecule was first observed under substrate-limiting conditions (60 nM ATP) to determine the 3 binding dwell angles. Then,

the buffer containing an inhibitor (1  $\mu M$  AMP-PNP or 1 mM  $NaN_3$ ) was introduced into the flow chamber with 60 nM ATP for the AMP-PNP inhibition or 200  $\mu M$  ATP for  $N_3^-$  inhibition. When the motor exhibited a long pause ( $>3$  min), it was verified that the pause was not caused by ADP inhibition by forcibly rotating the motor through  $+80^\circ$  with the magnetic tweezers; when inhibited by AMP-PNP or  $N_3^-$ ,  $F_1^{\alpha\beta\gamma}$  was never activated by this manipulation. However, when AMP-PNP- or  $N_3^-$ -inhibited  $F_1^{\alpha\beta\gamma}$  was forcibly rotated well over  $+80^\circ$ , for example  $+180^\circ$ , some molecules resumed rotation and stopped after a few turns. Such harsh manipulation would repel the tightly bound AMP-PNP- or  $N_3^-$ -ADP from  $F_1^{\alpha\beta\gamma}$ . After determining the pause position of inhibited  $F_1^{\alpha\beta\gamma}$ , the buffer was replaced with the inhibitor-free ATP buffer to confirm the molecule was still active by observing the resumption of ATP-driven rotation. Fig. 5A shows the pause position under AMP-PNP inhibition. The red bars represent the angle distribution during the  $120^\circ$  stepping rotation, and blue bars represent the angle distribution when inhibited by the inhibitor mixture. The angular distance ( $\Delta\theta$ ) of the AMP-PNP pause from the nearest binding angle on the clockwise side was determined (Fig. 5B) as in the above cross-link experiment. The mean angular distance was  $84.2^\circ \pm 16.3^\circ$ . This position corresponds to the catalytic dwell angle, consistent with the results of the cross-link experiment. The same result was obtained in the  $N_3^-$  inhibition experiments, and the mean angular distance of  $81.9^\circ \pm 16.3^\circ$  was obtained (Fig. 5 C and D). The pause angle of  $F_1^{\alpha\beta\gamma}$  inhibited by 200  $\mu M$  AMP-PNP, 5  $\mu M$  ADP, and 1 mM  $NaN_3$ , which mimics the crystallization buffer of the reference structure, was also examined. The mean angular distance was  $80.2^\circ \pm 14.5^\circ$  (Fig. 5 E and F), essentially the same as the above experiments. Thus, many lines of experiments confirmed that the crystal structure of  $F_1$  represents the catalytic dwell state found in the rotation assay.

## Discussion

The crystal structure of  $F_1$  was experimentally shown to represent the conformation of the catalytic dwell state found in single-molecule rotation assay. This means each  $\beta$ -subunit is in the  $\beta_{TP}$ ,  $\beta_{DP}$ , or  $\beta_{empty}$  conformations during the catalytic dwell state. The  $\beta$ - $\gamma$  cross-link using a chimera  $F_1^{\alpha\beta\gamma}$  carrying 1 copy of  $\beta(E190D/E391C)$  showed that the catalytic angle of  $\beta(E190D/E391C)$  coincides with the pause position of  $F_1^{\alpha\beta\gamma}$  where  $\beta_{DP}$  is cross-linked with  $\gamma$ . This finding means that the  $\beta_{DP}$  corresponds to the conformational state of the  $\beta$  that executes ATP hydrolysis reaction, consistent with theoretical works on quantum mechanics and molecular mechanics (31) or free-energy difference simulations (32). The crystal structure of  $BeF_3^-F_1$ , which is thought to mimic the catalytic intermediate state, also supports this result (22).

Fig. 6 shows the present reaction scheme of  $F_1$ , in which there are 2 chemical states during the catalytic dwell: prehydrolysis (state B) and posthydrolysis states (state C). Both states have the corresponding crystal structures. The recently reported structures (19, 33), the so-called “ground state” structures, in which both  $\beta_{TP}$  and  $\beta_{DP}$  bind with AMP-PNP correspond to the prehydrolysis state. However, the  $ADP \cdot AlF_3$ -bound structure (34) would represent the posthydrolysis state. The azide-bound structures (10, 18) would also correspond to this state. Considering that  $\beta_{DP}$  represents the state that hydrolyzes ATP,  $\beta_{DP}$  corresponds to the  $\beta$  before and after executing hydrolysis (the  $\beta$  at the right bottom in states B and C). Consequently,  $\beta_{TP}$  conformation represents the ATP-bound state that was the ATP-waiting state in the prior binding dwell state (state A in Fig. 6), and  $\beta_{empty}$  corresponds to the state after ADP release.

Thus, this study established the correlation between the conformational states of  $F_1$  found in the crystal structures and the single-molecule rotation assay. However, there is one uncertain point in the reaction scheme, the timing of  $P_i$  release. The present model assumes that  $\beta_{empty}$  is the  $P_i$ -bound state, and after  $\beta_{empty}$  releases  $P_i$ , the  $40^\circ$  substep is triggered (state C to state D). This assumption is based on the recent study of the crystal structure of

yeast MF<sub>1</sub> (33) in which P<sub>i</sub> binds to β<sub>empty</sub>. The P<sub>i</sub>-binding residues in the structure are consistent with those that biochemical studies have identified as the P<sub>i</sub> binding site (35), supporting our reaction model. However, considering the other crystal structures that do not show an obvious electron density of P<sub>i</sub> in the putative P<sub>i</sub>-binding site, an alternative model is still possible; P<sub>i</sub> is released from the β<sub>DP</sub> immediately after hydrolysis. A clarification of this issue is the next challenging task in the study of the mechanochemical-coupling mechanism of F<sub>1</sub>-ATPase.

Another issue is the structure of the binding dwell state. Although the ADP·AlF<sub>4</sub><sup>-</sup>-bound MF<sub>1</sub> shows obvious structural differences, the direction of the twist in the γ is opposite to our expectations. Therefore, ADP·AlF<sub>4</sub><sup>-</sup>-bound MF<sub>1</sub> is not in the binding dwell state. Rather, it might represent the intermediate state between state A and B of our reaction scheme in Fig. 6. Thus, there are no structural data about the binding dwell state. Because it is crucial for the understanding of the mechanochemical coupling mechanism of F<sub>1</sub>, a crystallographic study of F<sub>1</sub> in the binding dwell state is one of the most important tasks for the understanding of the mechanism of F<sub>1</sub>-ATPase. However, the fact that the crystal structures identified so far are comparable with the reference structure implies that the crystallization of F<sub>1</sub> in the binding dwell state is very difficult. One possible way is to crystallize F<sub>1</sub> at low temperature (16) or F<sub>1</sub> inhibited by tentoxin (36), where it spends most of the catalytic turnover time pausing at the binding dwell angle.

## Materials and Methods

**Preparation of F<sub>1</sub><sup>αβγ</sup>.** Throughout this work, α<sub>3</sub>β<sub>3</sub>γ subcomplex of F<sub>1</sub>-ATPase from the thermophilic *Bacillus* P53 (TF<sub>1</sub>) was used. For inhibition by AMP-PNP or N<sub>3</sub><sup>-</sup>, wild-type F<sub>1</sub><sup>αβγ</sup> modified for the rotation assay α(His<sub>56</sub> at N terminus/C193S)β(His<sub>10</sub> at N terminus)γ(S108C/I211C) was used. For simplicity, this F<sub>1</sub><sup>αβγ</sup> was referred to as wild-type F<sub>1</sub><sup>αβγ</sup> or α<sub>3</sub>β<sub>3</sub>γ. For the β-γ cross-link experiment, cysteine residues were

introduced into γR84 and/or βE391 of wild-type F<sub>1</sub><sup>αβγ</sup> and α<sub>3</sub>β(E190D)<sub>3</sub>γ (14) to construct 4 mutants: α<sub>3</sub>β<sub>3</sub>γ(R84C), α<sub>3</sub>β(E190D)<sub>3</sub>γ(R84C), α<sub>3</sub>β(E391C)<sub>3</sub>γ(R84C), and α<sub>3</sub>β(E190D/E391C)<sub>3</sub>γ(R84C). Mutagenesis to construct the expression vectors of these mutants was performed as per the previous report on β-γ cross-linking (25). The mutants of F<sub>1</sub><sup>αβγ</sup> were expressed in *E. coli*, purified, and biotinylated as reported (37). For the reconstitution of the chimera, α<sub>3</sub>β<sub>2</sub>β(E190D/E391C)γ(R84C), solutions of α<sub>3</sub>β<sub>3</sub>γ(R84C) and α<sub>3</sub>β(E190D/E391C)<sub>3</sub>γ(R84C) were mixed in a molar ratio of 2:1 and incubated for >2 days in the presence of 200 mM NaCl and 100 mM DTT at 4 °C and pH 7.0. The chimera α<sub>3</sub>β(E190D)<sub>2</sub>β(E391C)γ(R84C) was prepared by mixing solutions of α<sub>3</sub>β(E190D)<sub>3</sub>γ(R84C) and α<sub>3</sub>β(E391C)<sub>3</sub>γ(R84C) in a molar ratio of 2:1.

**Rotation Assay.** The rotary motion of F<sub>1</sub><sup>αβγ</sup> was visualized by attaching a magnetic bead (<0.2 μm; Seradyn) onto the γ-subunit of F<sub>1</sub><sup>αβγ</sup> and immobilizing the α<sub>3</sub>β<sub>3</sub> ring on a Ni-NTA-modified glass surface. Phase-contrast images of the rotating bead were obtained with an inverted optical microscope (IX-70; Olympus) equipped with magnetic tweezers (30). The image was captured with a CCD camera (FC300M; Takenaka) and recorded with a DV-CAM (DSR-11; Sony) at 30 fps. The recorded images were analyzed with image analysis software (Celery; Library) or a custom-made program (K. Adachi, Waseda University). The experimental procedures of the rotation assay for β-γ cross-linking were mostly the same as those reported (16), except for the content of the buffer. The basal buffer for the rotation assay contained 50 mM Hepes-KOH at pH 8.0, 50 mM KCl, 2 mM MgCl<sub>2</sub>, 1 mM phospho(enol)pyruvate, 0.1 mg/ml pyruvate kinase, 5 mg/ml BSA, and 1 mM DTT. For β-γ cross-linking, 200 μM DTNB was added to the basal buffer, from which DTT and BSA were omitted. The rotation assay for AMP-PNP and/or NaN<sub>3</sub> inhibition was carried out in the buffer containing 50 mM Mops-KOH at pH 7.0, 50 mM KCl, 2 mM MgCl<sub>2</sub>, 5 mg/ml BSA, and the indicated amount of nucleotides and NaN<sub>3</sub>.

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# Supporting Information

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## β subunit

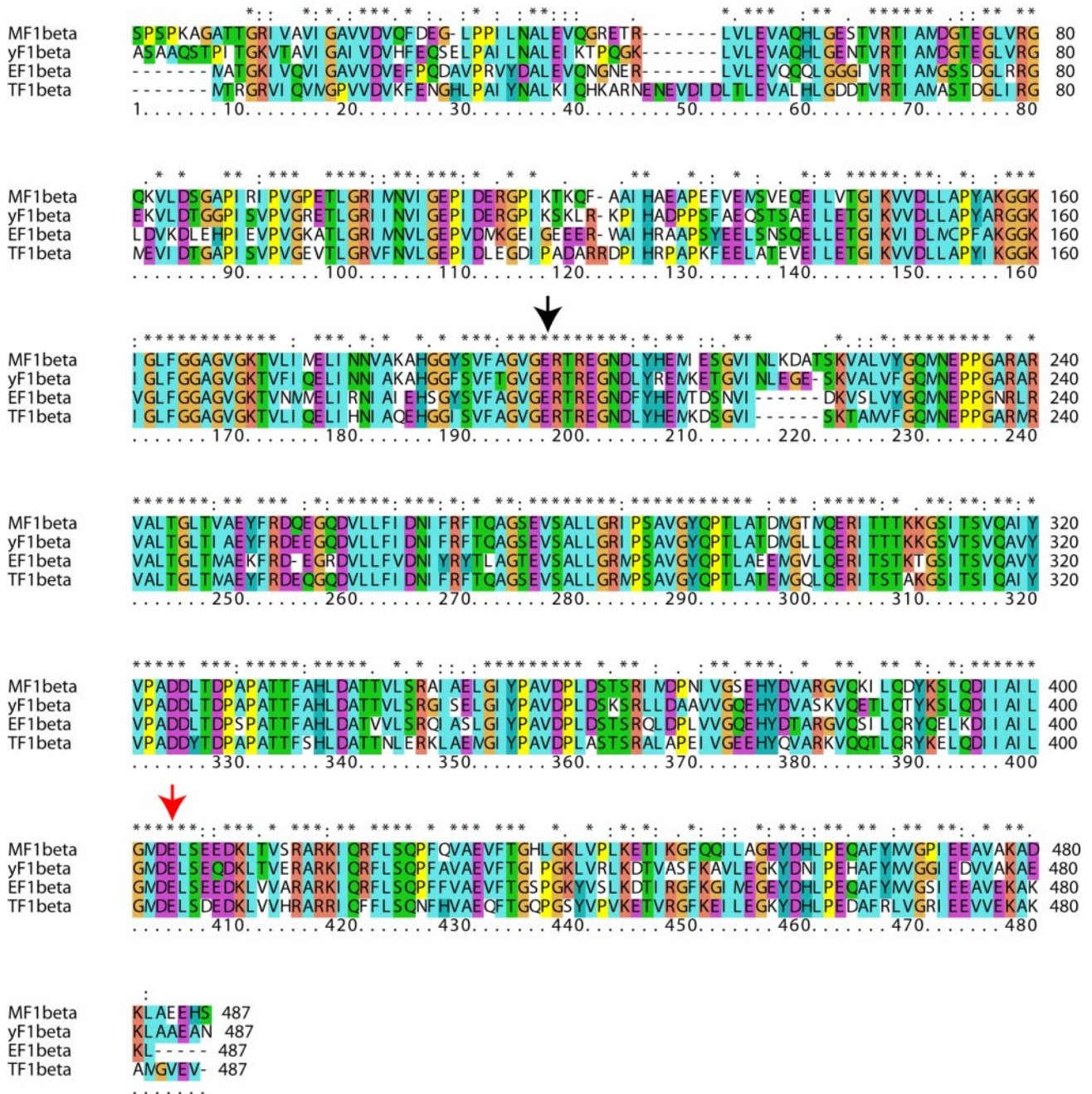
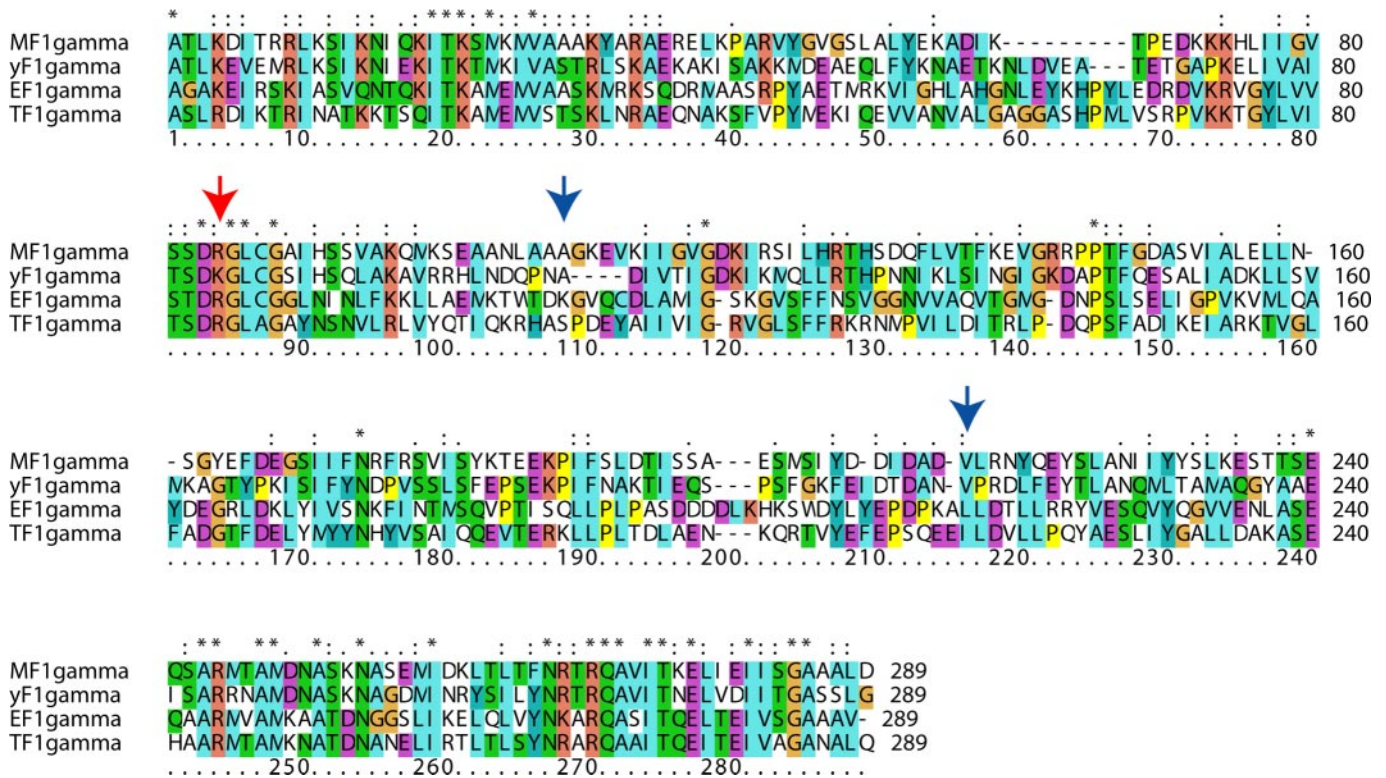


Fig. S1. Sequence alignment of the β-subunit of F<sub>1</sub> from bovine mitochondria (MF1), yeast mitochondria (yF1), *Escherichia coli* (EF1), and thermophilic *Bacillus* PS3 (TF1). A red arrow indicates the position of mutation into cysteine for cross-link. A black arrow indicates the mutated position of the βE190D. Sequence alignment was carried out by using ClustalX 2.0.9 software (1). Amino acid residues of precursor peptides were not included for alignment.

1. Larkin MA, et al. (2007) *Bioinformatics* 23:2947–2948.

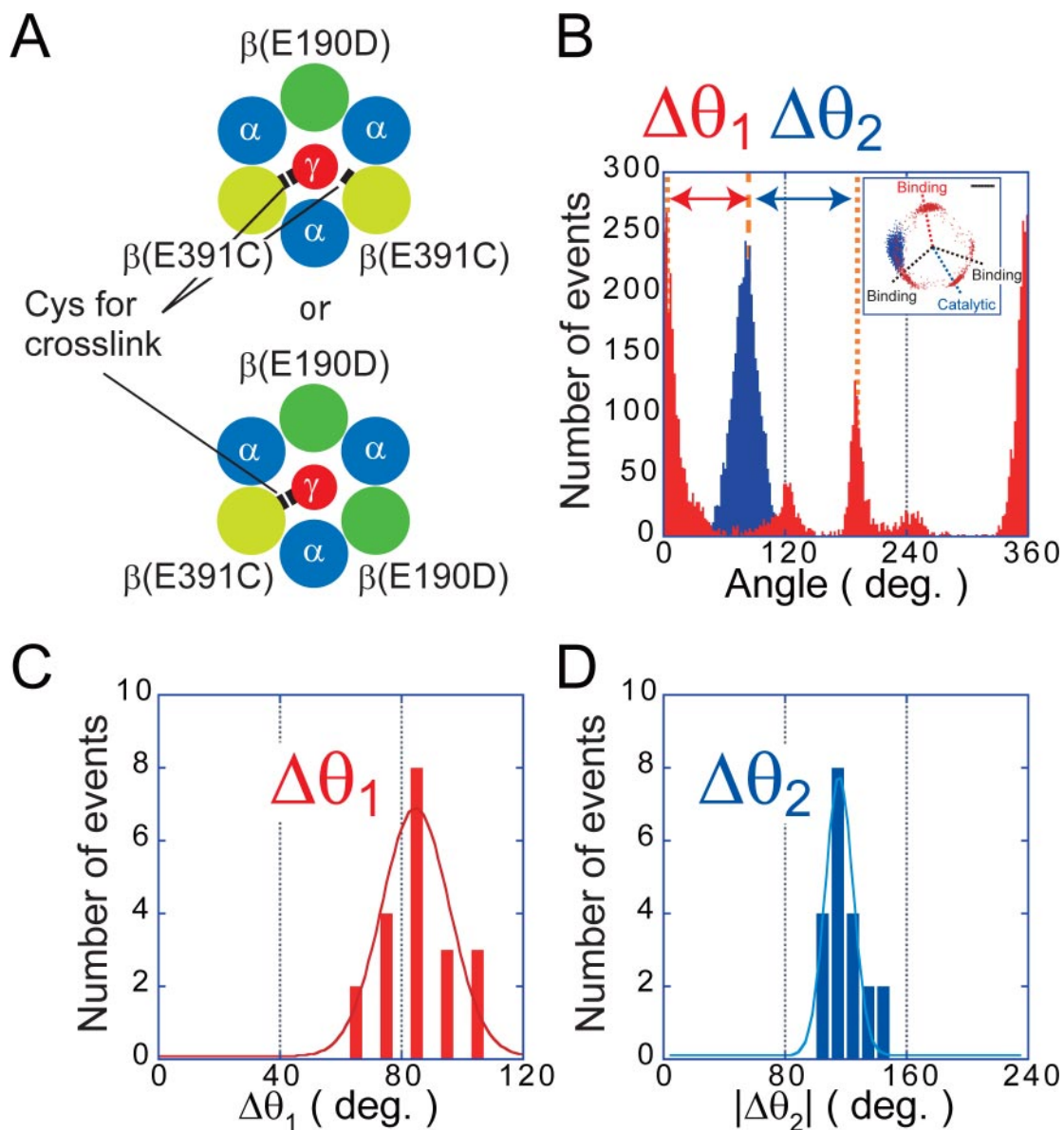
# $\gamma$ subunit



**Fig. S2.** Sequence alignment of the  $\gamma$ -subunit of MF<sub>1</sub>, yF<sub>1</sub>, EF<sub>1</sub>, and TF<sub>1</sub>. A red arrow indicates the position of mutation into cysteine for cross-link. Blue arrows indicate the mutated positions to cysteine for biotinylation. Sequence alignment was carried out by using ClustalX 2.0.9 software (1).

1. Larkin MA, et al. (2007) *Bioinformatics* 23:2947–2948.





**Fig. S3.** The pausing positions of chimera  $F_1^{\alpha\beta\gamma}$ ,  $\alpha_3\beta(\text{E190D})_2\beta(\text{E391C})\gamma(\text{R84C})$  or  $\alpha_3\beta(\text{E190D})\beta(\text{E391C})_2\gamma(\text{R84C})$ . (A) Schematic images of the chimera  $F_1^{\alpha\beta\gamma}$  molecules. (B) Typical angle distribution of a  $\alpha_3\beta(\text{E190D})\beta(\text{E391C})_2\gamma(\text{R84C})$  molecule. Red bars represent the angular position of free rotation before cross-link. Blue bars represent pausing angular position after cross-link. (Inset) The trace of centroid of the bead image. (Scale bar: 100 nm.) (C) Histogram of angle deviation of pausing position from the nearest binding dwell angle on the clockwise side ( $\Delta\theta_1$ ). (D) Histogram of angle deviation of pause position from the catalytic angle of  $\beta\text{E190D}$ . Because  $\alpha_3\beta(\text{E190D})\beta(\text{E391C})_2\gamma(\text{R84C})$  paused at two catalytic angles that correspond to those of wild-type  $\beta$ -subunits, in other words at the position about  $\pm 120^\circ$  from the catalytic angle of  $\beta\text{E190D}$ , distribution of absolute values is given. In C and D, data from two kind of chimera were combined. The total molecule and trial numbers were 11 [of them, two were  $\alpha_3\beta(\text{E190D})_2\beta(\text{E391C})\gamma(\text{R84C})$ ] and 20, respectively.