

## JB Review

# Rotation and structure of $F_0F_1$ -ATP synthase

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**$F_0F_1$ -ATP synthase is one of the most ubiquitous enzymes; it is found widely in the biological world, including the plasma membrane of bacteria, inner membrane of mitochondria and thylakoid membrane of chloroplasts. However, this enzyme has a unique mechanism of action: it is composed of two mechanical rotary motors, each driven by ATP hydrolysis or proton flux down the membrane potential of protons. The two molecular motors interconvert the chemical energy of ATP hydrolysis and proton electrochemical potential via the mechanical rotation of the rotary shaft. This unique energy transmission mechanism is not found in other biological systems. Although there are other similar man-made systems like hydroelectric generators,  $F_0F_1$ -ATP synthase operates on the nanometre scale and works with extremely high efficiency. Therefore, this enzyme has attracted significant attention in a wide variety of fields from bioenergetics and biophysics to chemistry, physics and nanoscience. This review summarizes the latest findings about the two motors of  $F_0F_1$ -ATP synthase as well as a brief historical background.**

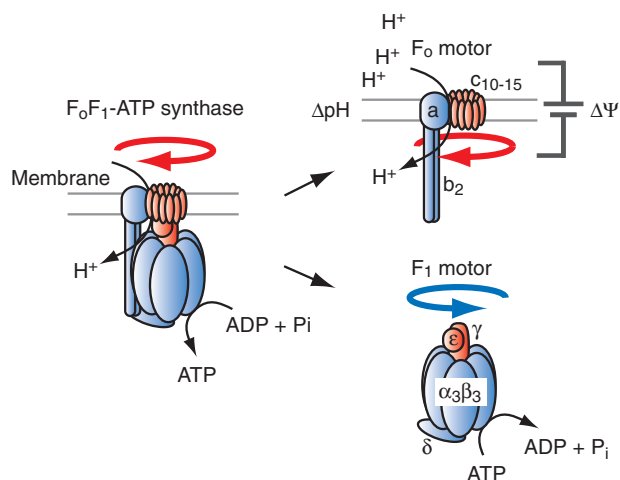
**Keywords:** ATP hydrolysis/ $F_0F_1$ -ATP synthase/high reversibility/rotary motor/stepping rotation.

**Abbreviations:** ADP, adenosine diphosphate; AMP-PNP, adenosine-5'-( $\beta,\gamma$ -imino)-triphosphate; ATP, Adenosine-5'-triphosphate; ATP $\gamma$ S, adenosine 5'-( $\gamma$ -thio)triphosphate;  $P_i$ , inorganic phosphate.

### $F_0F_1$ -ATP synthase

Adenosine-5'-triphosphate (ATP) is the ubiquitous energy currency of the cell. The human body contains about 50 g of ATP that is sustained by strict dynamic equilibrium between hydrolysis and synthesis. The total ATP produced under basal metabolism in humans amounts to 50–75 kg per day, and the same amount of ATP is consumed for the large variety of energy-requiring reactions such as muscle contraction, synthesis of biomolecules and mass transfer across biomembranes. Under aerobic conditions, the major ATP synthesis pathway is oxidative phosphorylation of which the terminal reaction is catalysed by  $F_0F_1$ -ATP synthase.

This enzyme is found widely in the biological world, including in thylakoid membranes, the mitochondrial inner membrane and the plasma membrane of bacteria. This enzyme catalyses ATP synthesis from adenosine diphosphate (ADP) and inorganic phosphate ( $P_i$ ) by using the electrochemical potential of protons (or sodium ions in some bacteria) across the membrane, i.e. it converts the electrochemical potential into its chemical form. This enzyme also functions in the reverse direction when the electrochemical potential becomes insufficient: it catalyses proton pumping to form an electrochemical potential to hydrolyse ATP into ADP and  $P_i$ .  $F_0F_1$ -ATP synthase is a super-complex enzyme with a molecular weight of >500 kDa and consists of two rotary motors. One is  $F_1$  (~380 kDa), which is the water-soluble part of ATP synthase. When isolated from the membrane portion, it acts as an ATP-driven motor: it rotates its inner subunit to hydrolyse ATP and is therefore termed  $F_1$ -ATPase. The other rotary motor of ATP synthase is  $F_0$  (~120 kDa), which is embedded in the membrane and generates rotary torque upon proton translocation that is driven by proton electrochemical potential (Fig. 1) (*1*). Bacterial  $F_1$  is composed of  $\alpha_3\beta_3\gamma\delta\epsilon$ -subunits. The three  $\alpha$ - and  $\beta$ -subunits form the hexameric stator ring in which the  $\alpha$ - and  $\beta$ -subunits are alternately arranged. The rotor shaft is the  $\gamma$ -subunit, which is accommodated in the central cavity of the  $\alpha_3\beta_3$ -ring. The  $\epsilon$ -subunit binds onto the protruding part of the  $\gamma$ -subunit and provides a connection between the rotor parts of  $F_1$  and  $F_0$ . The  $\epsilon$ -subunit acts as the endogenous inhibitor of  $F_1$  (*2–4*), by transforming the conformational state from the closed form to extended form that blocks the  $\gamma$  rotation due to steric hindrance (*5–8*). This inhibitory function is thought to be physiologically important to avoid ATP consumption (*9*). The  $\delta$ -subunit acts as a connector between  $F_1$  and  $F_0$  that connects the stator parts. Thus, the minimum complex of  $F_1$  as a motor is the  $\alpha_3\beta_3\gamma$  subcomplex. Catalytic reaction centres for ATP hydrolysis/synthesis reside at the three  $\alpha$ – $\beta$  interfaces, which are on the anticlockwise side of the  $\beta$ -subunit as indicated with red circles in Fig. 2A. The non-catalytic ATP-binding sites reside on the other  $\alpha/\beta$  interfaces. While the catalytic site is formed mainly with amino-acid residues from the  $\beta$ -subunit, the non-catalytic sites are primarily within the  $\alpha$ -subunit. Upon ATP hydrolysis on the catalytic sites,  $F_1$  rotates the  $\gamma$ -subunit in the anticlockwise direction viewed from the  $F_0$  side.  $F_0$  part consists of  $ab_2c_{10–15}$  subunits. The number of  $c$  subunits varies among species. For example, the copy number of the  $c$  subunit is eight in bovine mitochondria (*10*), 10 in yeast (*11*), *Escherichia coli* (*12*) and thermophilic *Bacillus* PS3 (*13*), 11 in *Ilyobacter tartaricus* (*14, 15*),

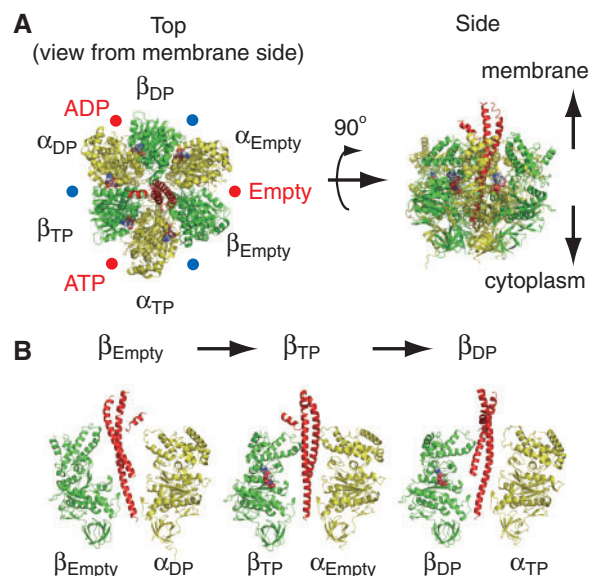


**Fig. 1**  $F_0$  and  $F_1$  motors of ATP synthase. Schematic images of  $F_0F_1$ -ATP synthase. The rotor and stator parts are shown in red and blue, respectively. The subunit composition of bacterial  $F_0$  is  $ab_2c_{10-15}$  (the number of c subunits varies from 10 to 15 in different species).  $F_0$  is embedded in the cell membrane and rotates the  $c$ -ring against the  $ab_2$  stator, driven by passive proton translocation along the proton electrochemical potential that comprises the proton concentration ( $\Delta p\text{H}$ ) and membrane voltage ( $\Delta \Psi$ ) across the membrane. Bacterial  $F_1$  is composed of  $\alpha_3\beta_3\gamma\delta\epsilon$  and is an ATP-driven rotary motor in which the  $\gamma$ -subunit rotates against the  $\alpha_3\beta_3$ -cylinder. The  $\epsilon$ -subunit binds to the protruding part of the  $\gamma$ -subunit. The  $\delta$  binds to the bottom of the  $\alpha_3\beta_3$ -ring (note that the rotational direction of  $F_0$  is opposite to that of  $F_1$ ). In the whole complex of  $F_0F_1$ ,  $F_0$  reverses the rotation of  $F_1$ , leading to ATP synthesis from ADP and  $P_i$ .

*Propionigenium modestum* (16) and *Clostridium parapdxum* (17), 13 in thermoalkaliphilic *Bacillus* TA2.TA1 (18) and *Bacillus pseudofirmus* OF4 (19), 14 in spinach chloroplast (20) and 15 in *Spirulina platensis* (21). The  $c$  subunits form a ring complex by aligning in a circle. It is widely thought that the  $c$ -ring and the  $a$  subunit form a proton pathway (for details, see the 'Proton translocation pathway of  $F_0$ ' section). With the downhill proton flow through the proton channel, the  $c$ -ring rotates against the  $ab_2$  subunits in the opposite direction of the  $\gamma$ -subunit of the  $F_1$  motor (22). Thus, in the  $F_0F_1$  complex,  $F_0$  and  $F_1$  push each other in the opposite direction. Under physiological condition where the electrochemical potential of the protons is large enough to surpass the free energy of ATP hydrolysis,  $F_0$  forcibly rotates the  $\gamma$ -subunit in the clockwise direction and then  $F_1$  catalyses the reverse reaction, i.e. ATP synthesis which is the principle physiological function of ATP synthase. In contrast, when the electrochemical potential is small or decreases,  $F_1$  forces  $F_0$  to rotate the  $c$ -ring in the reverse direction to pump protons against the electrochemical potential.

### Binding change mechanism and structure of $F_1$ -ATPase

The three catalytic sites on the  $\beta$ -subunits work cooperatively during catalysis. The classic working model for  $F_1$  is the 'binding-change mechanism' proposed by Paul Boyer (23). The early stage of this model



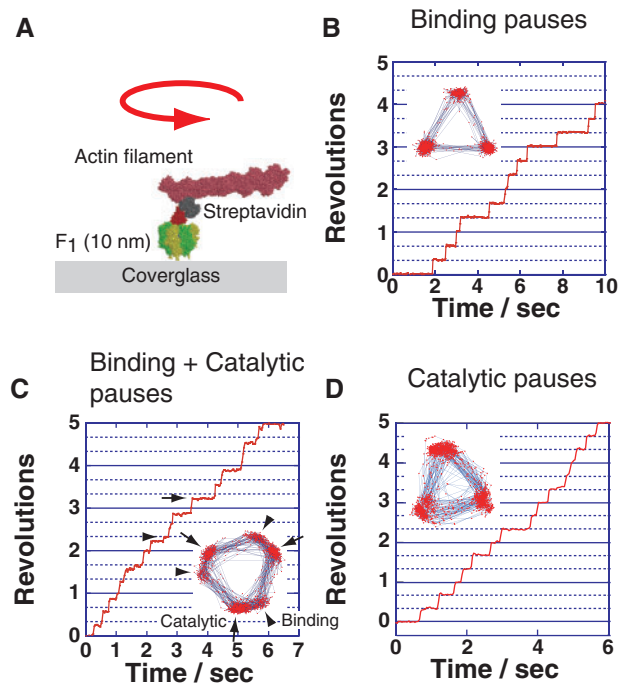
**Fig. 2** Crystal structure of the  $\alpha_3\beta_3\gamma$  subcomplex of  $F_1$ . The crystal structure of  $F_1$  from bovine mitochondria (PDB code; 1BMF). The  $\alpha$ -,  $\beta$ - and  $\gamma$ -subunits are shown in yellow, green and red, respectively. (A) The left figure is viewed from the membrane side ( $F_0$  side), and is rotated 90° in anticlockwise direction to arrow (right figure). The protruding part of  $\gamma$  is directed toward the membrane side (15). The catalytic sites are located at the  $\alpha$ - $\beta$  interface indicated by red circles, which are primarily on the  $\beta$ -subunit. Each site carries AMP-PNP, ADP, or is empty and is designated as  $\beta_{TP}$ ,  $\beta_{DP}$ , or  $\beta_{Empty}$ , respectively. The other interfaces are non-catalytic sites (blue circles), all of which bind with AMP-PNP. Each  $\alpha$ -subunit forming a catalytic site with the  $\beta$ -subunit is designated as  $\alpha_{TP}$ ,  $\alpha_{DP}$  and  $\alpha_{Empty}$ , respectively. (B) Conformational states of the  $\beta$ -subunit and the catalytic  $\alpha$ - $\beta$  interfaces. Three  $\alpha$ - $\beta$  pairs with the  $\gamma$ -subunit are shown in yellow and green with the central  $\gamma$ -subunit (red). The  $\alpha$  and  $\beta$ -subunits are composed of the N-terminal domain, nucleotide-binding domain and C-terminal domain (from bottom to top).  $\beta_{Empty}$  has an open conformation in which the  $\alpha$ -helical C-terminal domain rotates upwards to open the cleft of the nucleotide-binding pocket. Both  $\beta_{ATP}$  and  $\beta_{ADP}$  have a closed conformation entrapping the nucleotide within the closed pocket.

postulated alternating transition between two chemical states, assuming two catalytic sites residing on  $F_1$ . It was later revised to propose the cyclic transition of the catalytic states among three catalytic sites based on the biochemical and electron microscopic experiments that revealed that  $F_1$  has the three catalytic sites (24–26). One important feature of this model is that the affinity for nucleotide at each catalytic site is different from each other at any given time, and the status of the three  $\beta$ -subunits cooperatively change in one direction accompanying  $\gamma$  rotation. This hypothesis is strongly supported by X-ray crystallographic studies performed by Walker's group (27). The first resolved crystal structure of  $F_1$  (27) revealed many essential structural features of  $F_1$  at atomic resolution. Importantly, the catalytic  $\beta$ -subunits differ from each other in conformation and catalytic state: one binds to an ATP analogue, adenosine-5'-( $\beta,\gamma$ -imino)-triphosphate (AMP-PNP), the second binds to ADP and the third site is empty (Fig. 2A). Therefore, these sites are termed  $\beta_{TP}$ ,  $\beta_{DP}$  and  $\beta_{Empty}$ , respectively. While  $\beta_{TP}$  and  $\beta_{DP}$  have a closed conformation wrapping bound nucleotides on the catalytic sites,  $\beta_{Empty}$  has an open conformation

swinging the C-terminal domain away from the binding site to open the cleft of the catalytic site (Fig. 2B). These features are consistent with the binding-change mechanism. Another important feature found in the crystal structure is that while the N-terminal domains of the  $\alpha$  and  $\beta$ -subunits form a symmetrical smooth cavity as the bearing for  $\gamma$  rotation at the bottom of the  $\alpha_3\beta_3$ -ring, the C-terminal domains of the  $\beta$ -subunit show distinct asymmetric interactions with the  $\gamma$ -subunit. Therefore, the most feasible inference is that the open-to-closed transition of the  $\beta$ -subunits upon ATP binding pushes  $\gamma$ , and the sequential conformational change among  $\beta$ -subunits leads the unidirectional  $\gamma$  rotation, which was recently visualized in simultaneous imaging of the conformational change of the  $\beta$ -subunit and the  $\gamma$  rotation (28).

### Verification of $F_1$ rotation by single-molecule observation

Since the publication of the crystal structure, many studies have attempted to demonstrate the rotation of  $F_1$ . Crosslink exchange experiment between the  $\beta$ - and  $\gamma$ -subunits of  $F_1$  derived from *E. coli* (29) and the polarized absorption relaxation of  $F_1$  from spinach chloroplasts (30) have proven the rotational motion of the  $\gamma$ -subunit during catalysis. Unidirectional rotation of the  $\gamma$ -subunit upon ATP hydrolysis was proved with the direct observation of  $F_1$  rotation from thermophilic *Bacillus* PS3 ( $TF_1$ ) under the microscope (31). In order to suppress rotary Brownian motion study,  $F_1$  was immobilized on a glass surface modified with Ni-nitrilotriacetic acid (NTA) thorough the interaction between  $Ni^{2+}$  and the His-tag, which was introduced into the N-terminus of the  $\beta$ -subunit. In addition, a fluorescently labelled actin filament with length of 0.6–4  $\mu\text{m}$  and diameter of 10 nm was attached to the  $\gamma$ -subunit as the rotation marker to magnify the subtle motion of the  $\gamma$ -subunit of which the radius is only 1 nm, which is much smaller than the spatial resolution ( $\sim 200$  nm) of a conventional microscope (Fig. 3A). Note that in recent studies, other types of probes such as polystyrene beads, gold colloidal beads, gold nanorods, and magnetic beads are frequently used instead of actin filaments because the imaging of fluorescently labelled actin filaments suffers from photobleaching. The rotational direction is always anticlockwise when viewed from the  $F_0$  side and, importantly, it was consistent with the expected rotary direction from the crystal structure. Assuming the  $\beta$ -subunit undergoes the conformational transition from  $\beta_{\text{Empty}}$ ,  $\beta_{\text{TP}}$  and  $\beta_{\text{DP}}$ , each catalytic state propagates in the anticlockwise direction, accompanying the anticlockwise  $\gamma$  rotation. The rotational velocity was far slower than the expected rate from bulk ATPase measurements because of the large hydrodynamic friction exerted on the rotating actin filament. However, this allows us to estimate the torque generated by individual  $F_1$  molecules from the hydrodynamic friction that should be in equilibrium with  $F_1$ 's torque. The torque was determined to be around 40 pN nm. Although this is a rough estimation without



**Fig. 3 Single-molecule rotation assay of  $F_1$ .** (A) A schematic image of the experimental setup. The  $\alpha_3\beta_3$ -ring is fixed on the glass surface to suppress translational and rotational Brownian motion of the  $F_1$  molecule. A rotation probe (fluorescently-labelled actin filament) is attached to the  $\gamma$ -subunit to visualize the rotary motion under an optical microscope. (B) Rotation of  $F_1$ -ATPase under ATP-limiting conditions (60 nM ATP). Inset shows the trajectory of the centroid of the probe. (C) Rotation of mutant  $F_1$ -ATPase,  $\beta(E190D)$ , at 2  $\mu\text{M}$  ATP. Under this condition,  $120^\circ$  step is divided into  $0^\circ$  and  $80^\circ$  dwelling positions. Each pause corresponds to ATP binding and ATP catalytic dwelling positions, respectively. Arrow heads and arrows indicate the positions of ATP binding and catalytic dwell, respectively. (D) Rotation of a mutant  $F_1$ -ATPase,  $\beta(E190D)$ , at saturating ATP (2 mM). Hydrolysis rate is slowed by the mutation so that three pauses to wait for the hydrolysis reaction are observed.

consideration of the viscosity increment in the immediate vicinity of surface, the value was recently confirmed to be valid using more precise torque measurements based on fluctuation theorem, which estimates the entropy generation upon the rotation without assuming the friction coefficient (32). Taking into account that the step size is  $120^\circ$ , each coupled with single ATP hydrolysis turnover as below,  $F_1$  works with 80 pN nm, which corresponds to the free energy released from hydrolysis of a single ATP molecule under physiological conditions, suggesting high 100% energy conversion efficiency of  $F_1$ .

### Stepping rotation of $F_1$

Many attempts have been made to resolve rotary motion into discrete steps to clarify how the rotation is coupled with each elementary catalytic step of ATP hydrolysis: ATP-binding, hydrolysis and product release. The stepping rotation was first observed in the rotation assay with actin filaments under ATP-limiting conditions, where the ATP-binding process determines the net turnover rate of ATP hydrolysis and rotation.

When [ATP] is well below the Michaelis–Menten constant ( $K_M$ ) of the rotation ( $\sim 1 \mu\text{M}$ ),  $F_1$  showed discrete  $120^\circ$  steps that were intervened with pauses, consistent with the pseudo 3-fold symmetry of the  $\alpha_3\beta_3$ -ring (Fig. 3B). The mean dwell time of the pause before the steps was inversely proportional to [ATP], suggesting that each step is triggered by a single event of ATP binding. A histogram of the dwell time showed an exponential decay with the time constant in consistent with the observed mean dwell time, implying that the single event triggers the  $120^\circ$  step (33). The coupling ratio of a single  $120^\circ$  step per ATP was directly confirmed in a later study (34). However, the stepping rotation was not detected at ATP-saturating conditions owing to damping by high viscous friction against actin filaments. Therefore, a very small probe was employed to detect the intrinsic stepping motion of  $F_1$ . A single gold colloid (40 nm) was attached to the  $\gamma$ -subunit so that viscous friction was negligible, and the maximum rotational velocity reached and exceeded the expected rate from bulk ATPase (35). The discrepancy from bulk ATPase is attributed to some fraction of  $F_1$  being in an inactive state, the so-called ADP-inhibited form (36), in the ensemble measurement. In this rotation assay, the  $120^\circ$  step rotation was observed even under ATP-saturating conditions. Near the  $K_M$ , where time constants for the ATP-binding step and other catalytic steps are comparable, the rotation showed two substeps of which angular displacement were resolved into  $90^\circ$  and  $30^\circ$  (35). In a following experiment, in order to facilitate the analysis of the catalytic dwell, a mutation was introduced at the catalytic site,  $\beta\text{E190D}$  (thermophilic *Bacillus* PS3) that significantly slows the rate constant of hydrolysis step (37). Around  $K_M$ , the mutant  $F_1$  shows six pauses composed of  $0^\circ$  and  $80^\circ$  dwelling positions during rotation, revising the substep sizes to be  $80^\circ$  and  $40^\circ$  (Fig. 3C). Kinetic analyses of the dwell time at  $0^\circ$  and  $80^\circ$  dwelling positions revealed that these substeps are triggered by ATP binding and two consecutive reactions with time constants around 1 ms, respectively. Recent studies have revised the two time constants at  $80^\circ$  dwelling position to be 1.3 ms and  $\sim 0.1$ – $0.3$  ms (38, 39). One of the two reactions at  $80^\circ$  dwelling position was revealed to be the hydrolysis step in the experiment that employed the aforementioned mutant  $F_1$  with slow hydrolysis rate and a slowly hydrolysing ATP analogue adenosine 5'-( $\gamma$ -thio)triphosphate (ATP $\gamma\text{S}$ ) (37). The angular dwelling positions at  $0^\circ$  and  $80^\circ$  are, therefore, termed the binding angle and catalytic angle (Fig. 3C and D), respectively. The angular positions of product release were investigated by adding an excess of ADP or  $\text{P}_i$  in the solution (38, 40). In the presence of ADP, the rotation was slowed because of lengthening of the dwell time at the binding angle, suggesting that the ADP-releasing angle is at a binding angle. Simultaneous imaging of fluorescently labelled nucleotide with the  $\gamma$  rotation also verified this point: fluorescently labelled ATP is released presumably as ADP after the  $\gamma$ -subunit rotates  $240^\circ$  or more from the angle where the nucleotide is bound to  $F_1$ . In contrast, in the presence of  $\text{P}_i$ ,  $F_1$  showed longer pauses at the catalytic

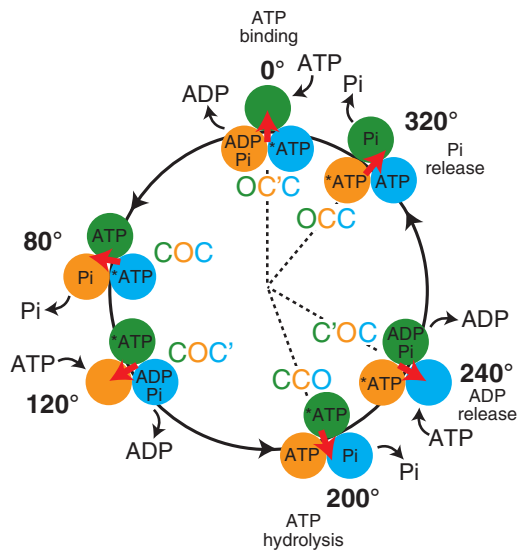
angle. Thus, it is thought that the release of ADP and  $\text{P}_i$  occurs at the binding and catalytic angles, respectively. Another intermediate of  $F_1$  at the binding angle was unexpectedly found in the rotation assay at low temperature,  $\sim 4^\circ\text{C}$  (40). This reaction showed an extremely high  $Q_{10}$  factor of 19, so this reaction is termed the temperature-sensitive reaction (TS). A direct correlation between TS and the ATP-binding or ADP-release step was not found although TS takes place at the binding. Considering the high  $Q_{10}$  factor, TS reaction might be some conformational rearrangement before or after ATP binding (41).

### Reaction scheme of $F_1$ -ATPase

As mentioned above, all of the elementary reaction steps were identified to occur at the binding angle or catalytic angle. However, because there are three positions for binding and catalytic angles, it is required for the establishment of the reaction scheme of  $F_1$  to determine at which angle each reaction occurs. ADP release was shown to occur at  $240$ – $320^\circ$ , but most likely at  $240^\circ$ . The angle for hydrolysis was determined using a hybrid  $F_1$  carrying a single copy of the aforementioned mutant  $\beta$ -subunit,  $\beta\text{E190D}$  (42). This hybrid allows us to identify the hydrolysis angle because the incorporated mutant  $\beta$ -subunit shows distinctly long pauses at two positions. One is at the ATP-binding angle of the mutant  $\beta$ -subunit ( $0^\circ$ ), and the other one is at  $+200^\circ$  from the binding angle. Thus, the hydrolysis angle was determined to be  $200^\circ$ . Note that the pause at  $0^\circ$  is due to the TS intermediate state (41), although it was attributed to ATP waiting dwell in the original report (43). The TS dwell could be confounded as the  $320^\circ$  pause presumably due to experimental error. The timing of  $\text{P}_i$  release has recently been determined to be at  $320^\circ$  in another type of experiment (39) where  $F_1$  was stalled with magnetic tweezers during hydrolysis dwell that was lengthened by  $\beta\text{E190D}$  and/or ATP $\gamma\text{S}$ . On the basis of the observation that bound ATP or ATP $\gamma\text{S}$  undergoes hydrolysis and synthesis in a reversible manner, it was shown that  $\text{P}_i$  (or thiophosphate) is not released immediately after hydrolysis at  $200^\circ$ . Because the  $\text{P}_i$  release has to be after hydrolysis and at a catalytic angle, it was concluded that  $\text{P}_i$  release occurs at  $320^\circ$ . Thus, the present reaction scheme of rotation and catalysis is as follows: ATP binding at  $0^\circ$ , hydrolysis at  $200^\circ$ , ADP release  $240^\circ$  and  $\text{P}_i$  release at  $320^\circ$  (Fig. 4).

### Correlation of reaction scheme with crystal structure

While the single-molecule rotation assay revealed that  $F_1$  has two stable conformations in pausing at the binding or catalytic angle, current crystal structures show essentially a single conformation. Correlation with the crystal structure remained obscure, although the interpretation of the crystal structure is crucial, especially for theoretical studies. Therefore, attempts have been made to determine whether the crystal structure represents the binding dwell or catalytic dwell



**Fig. 4 Mechanochemical coupling scheme of  $F_1$ .** Each circle represents the chemical state of the catalytic sites on the  $\beta$ -subunit. The red arrow represents the angular position of the  $\gamma$ -subunit. O, C' and C indicate the open, half-closed and closed forms, respectively. The green catalytic site retains the bound nucleotide as ATP until the  $\gamma$ -subunit rotates  $200^\circ$  from the binding angle ( $0^\circ$ ). At  $200^\circ$ , the catalytic site hydrolyses ATP into ADP and  $P_i$ , each of which is released at  $240^\circ$  and  $320^\circ$ , respectively. The conformation of the  $\beta$ -subunit changes from open to closed upon ATP binding and remains in the closed form until the  $\gamma$ -subunit rotates  $240^\circ$ , this  $\beta$ -subunit moves to the half-closed form, and then it returns to the open form with accompanying rotation of the  $\gamma$ -subunit.

(28, 44, 45). On the basis of the crystal structure, the characteristic interaction with the  $\gamma$ -subunit was identified in the  $\beta_{DP}$  form. Cysteine residues were genetically incorporated into the residues involved in the direct  $\beta$ - $\gamma$  contact. The mutant  $F_1$  was analysed in the rotation assay. During observation, incorporated cysteine residues were cross-linked through disulphide bonds by infusing an oxidizing buffer to stall  $F_1$  in the crystal structure form. The pausing angle corresponded to the catalytic angle. Thus, it was shown that the crystal structure represents the catalytic dwelling state and that  $\beta_{TP}$ ,  $\beta_{DP}$  and  $\beta_{Empty}$  correspond to the  $80^\circ$ ,  $200^\circ$  and  $320^\circ$  state, respectively (Fig. 4, green circles). These data were supported by experiments by Masaike *et al.* where the C-terminal domain of the  $\beta$ -subunit that undergoes the large swing motion upon ATP binding was labelled with a fluorescent dye (28). The observed angular positions in the catalytic dwell corresponded to those observed in the crystal structure. Interestingly, they found that at  $240^\circ$ , the  $\beta$ -subunit takes a new conformation, which they termed 'half-closed', while at other binding angles, the  $\beta$ -subunit takes the same conformation as the catalytic angles: open at  $0^\circ$  and closed at  $120^\circ$  (Fig. 4).

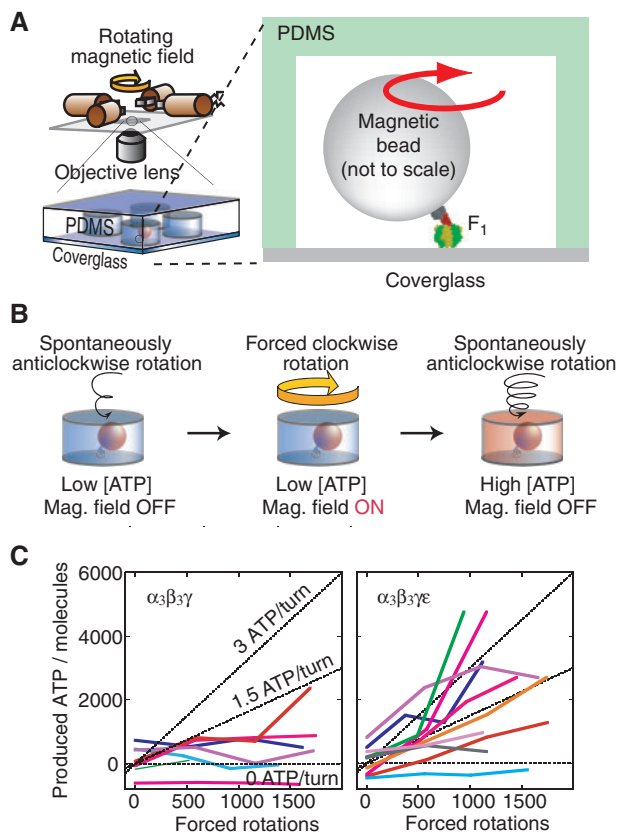
### ATP synthesis upon reverse rotation of $F_1$

Although the essential properties and basic mechanochemical coupling scheme of  $F_1$  as an ATP-driven motor have been established, the physiological role of  $F_0F_1$ -ATP synthase, that is ATP synthesis, has not

been sufficiently studied in single-molecule experiments. If ATP synthesis is a simple reverse reaction of hydrolysis, forcibly reversing rotation of  $F_1$  should lead to efficient ATP synthesis. Two lines of single-molecule experiments have been carried out to investigate this hypothesis. In the first experiment (46), a large number of  $F_1$  molecules were enclosed in an observation chamber and forcibly rotated in the reverse direction with a magnetic bead tweezer system. The synthesized ATP was detected as bioluminescence using the luciferin–luciferase system. Although ATP synthesis upon reverse rotation was clearly demonstrated, the uncertainty of the number of active  $F_1$  molecules in the chamber did not allow a quantitative estimation of the mechanochemical coupling ratio. Therefore, the following experiment focused a single active  $F_1$  molecule to determine the coupling ratio (34). The technical issue to be addressed was detection of a very small number of ATP molecules generated from a single  $F_1$  molecule. Even if we assume that  $F_1$  synthesizes three ATP molecules per one revolution at 10 Hz for 1 min, the total number of ATP molecule is only 1,800 molecules ( $3.0 \times 10^{-21}$  mol), which is far below the detection limit of the luciferase assay. To address this issue, a microscopic reaction chamber system was developed using a microfabrication technique, which has identically shaped reaction chambers, each of which is a few microns in scale and has a volume of 6 fL (47). Because the extremely small reaction volume resulted in high concentration, it was possible to detect a small amount of reaction product yielded from a single enzyme molecule. A single  $F_1$  molecule was encapsulated in the microchamber to accumulate synthesized ATP molecules (Fig. 5A and B). After forcible reverse rotation with magnetic tweezers,  $F_1$  was released from the tweezers. Because the rotational rate of ATP-driven rotation is proportional to [ATP] under the experimental conditions, one can measure the increment of [ATP] as that of the ATP-driven rotation rate. It was found that while the  $\alpha_3\beta_3\gamma$  subcomplex showed very weak ATP synthesis activity, the  $\alpha_3\beta_3\gamma\epsilon$  subcomplex had highly efficient ATP synthesis, up to  $\sim 80\%$  (2.3 ATP molecules per turn) (Fig. 5C). It is likely that the  $\epsilon$ -subunit stabilizes the protruding portion of the  $\gamma$ -subunit, as seen in the crystal structure, to transmit the applied torque to  $\gamma$ . This result implies that the efficiency of the mechanochemical coupling in ATP synthesis is also high in the whole  $F_0F_1$  complex. High reversibility of mechanochemical coupling is a remarkable feature of the ATP synthase that distinguishes it from other molecular motors; other motor proteins such as kinesin and myosin do not synthesize ATP when the movements are reversed by external force.

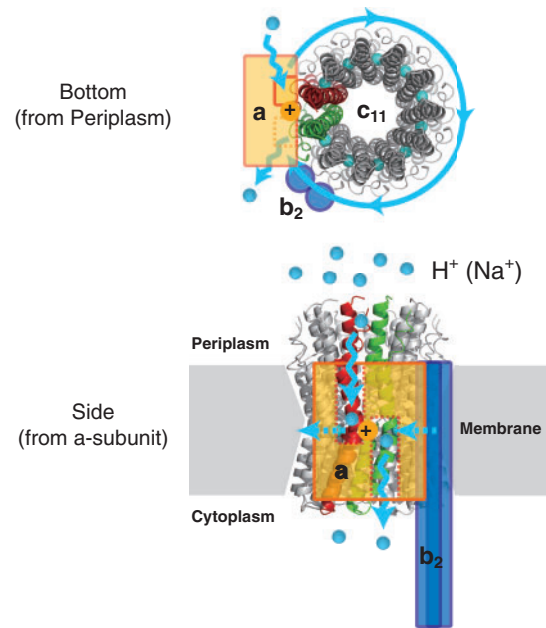
### Structure of $F_0$

Bacterial  $F_0$  has the common and simple subunit stoichiometry of  $ab_2c_{10-15}$ , while mitochondrial  $F_0$  has additional subunits: *d*, *e*, *f*, *OSCP*, *F6* and *A6L* (48). We hereafter focus on the minimum subcomplex of  $F_0$ ,  $ab_2c_{10-15}$ . The structure of the *c* subunit was first resolved in a monomer state by NMR (49). The *c*



**Fig. 5 ATP synthesis by reversing  $F_1$ .** (A) Schematic drawing of experimental setup. (B) Experimental procedure of ATP synthesis. Active single  $F_1$  is enclosed in a femtolitre chamber (left). A magnetic bead attached to the  $\gamma$ -subunit is forcibly rotated by magnetic tweezers (centre). Newly synthesized ATP is accumulated in the chamber. The number of synthesized ATP molecules is determined from the increments in the ATP-driven rotational speed of released  $F_1$  (right). (C) ATP synthesis by reversing  $\alpha_3\beta_3\gamma$  (left) and  $\alpha_3\beta_3\gamma\epsilon$  (right). Each trace is derived from individual  $F_1$  molecules. Dotted lines indicate slopes of the coupling ratio of 0% (0 ATP/turn), 50% (1.5 ATP/turn) and 100% (3 ATP/turn).

subunit has a hairpin structure that is composed of two  $\alpha$ -helices and a connecting loop. Crystal structures of the  $F_1$ - $c$ -ring structure (10, 11) and isolated  $c$ -ring (14–17, 19–21) revealed that the  $c$  subunits form a ring complex by assembling in a circle with the C-terminus pointing outwards and the connecting loop towards the  $F_1$  side (cytoplasmic side in bacteria) (Fig. 6). The  $b$  subunit has an N-terminal transmembrane domain. The  $b$  subunit forms a homodimer (50) which functions as the peripheral stalk to hold the stator parts of  $F_1$  and  $F_0$  to avoid slippage. It was shown that the  $b$  subunit has robustness against extensive deletion and insertion at the cytoplasmic helix (51, 52), suggesting that the  $b_2$  dimer also acts as the elastic connector for smooth torque transmission. The structure of the  $a$  subunit remains unclear. It is thought that this subunit has five transmembrane helices (53–55). Because the cavity of the  $c$ -ring is too small to accommodate the  $a$  subunit and/or the  $b_2$  dimer, each of which has five or two helices, it is reasonable to assume that the  $ab_2$  complex resides outside of the  $c$ -ring (56, 57).



**Fig. 6 Structure of  $F_0$ .** The  $c_{10}$ -ring is from the crystal structure of  $\text{Na}^+$ -transport  $F_0$  from *Ilyobacter tartaricus* (PDB code; 1YCE). The blue spheres at the middle of the  $c_{10}$ -ring represent bound  $\text{Na}^+$  ions. Schematic image of stator  $ab_2$  complex (thin orange and blue) is constructed based on the two-channel model. One half channel is exposed to the periplasmic side and the other to cytoplasmic side. Rotation of the  $c$ -ring accompanies the proton-transfer between the  $a$  and  $c$  subunits. Two  $c$  subunit monomers at the interface of the  $a$  subunit are shown in red and green, respectively. The shortcut transfer of protons between the  $c$  subunits connected to the half channels are blocked with the positive charge of the conserved Arg residue of the  $a$  subunit (depicted as the plus symbol).

### Proton translocation pathway of $F_0$ : the 2-channel model

The mechanical rotation of the  $c$ -ring by  $F_0$  is driven by proton flow through  $F_0$ . Although the structural basis of the proton translocation pathway is unknown, extensive biochemical work on  $F_0$  subunits has identified several charged residues in the transmembrane helices of the  $a$  and  $c$  subunits that would be directly involved in proton translocation. Among them, Asp or Glu of the  $c$  subunit and Arg of the  $a$  subunit, which correspond to  $c\text{Asp61}$  and  $a\text{Arg210}$  of *E. coli*  $F_0$ , are highly conserved among species and thought to have crucial roles in proton translocation. The crystal structure of the  $c$ -ring showed that the Asp residue (Glu in *I. tartaricus*  $F_0$ ) of the  $c$  subunit resides at the middle of the C-terminal helix. The recent structure of the  $c_{10}$ -ring from *I. tartaricus*  $F_0$ , which is a  $\text{Na}^+$ -transporting  $F_0$ , revealed that the Glu residues are occupied with  $\text{Na}^+$  ions (Fig. 6). Thus, it is well established that this conserved carboxyl residue is one of the proton-binding sites. However, other charged residues are not found in the  $c$  subunit in the vicinity of the carboxyl residue, suggesting that the  $a$  subunit has proton translocation pathways. The most widely accepted model on proton translocation in  $F_0$  is the so-called two-channel model, which assumes that the  $a$  subunit possesses two proton pathways each of which spans half of the membrane, but towards different sides;

the channels connect the proton-binding site of the  $c$  subunit with the periplasmic or cytoplasmic space (58–60) (Fig. 6). Notice each channel has contact with a different  $c$  subunit, which are adjacent to each other. In other words, the  $a$  subunit interacts with two  $c$  subunits, each contacting *via* a different half channel. The proposed mechanism of proton transfer in ATP synthesis mode is as follows (60–62): a proton enters the half channel exposed to the periplasmic side (or intermembrane space of mitochondria) and is then transferred to the carboxy residue of the  $c$  subunit. This protonation neutralizes the negative charge of the residue, allowing the  $c$  subunit to rotate apart from the  $a$  subunit towards the surrounding lipid layer. At the same time, the neighbouring  $c$  subunit at the anticlockwise side returns from the lipid layer to form contacts with the other half channel, which has a hydrophilic environment to promote deprotonation of the carboxyl residue. The released proton then enters into the cytoplasmic space. The role of the conserved Arg in the  $a$  subunit is likely to block the futile rotation of the  $c$  subunit without deprotonation by attracting only the deprotonated  $c$  subunit with its positive charge (62, 63). In the ATP-driven proton-pumping mode, the sequence of events is reversed.

### Rotation of $c$ -ring in $F_0$

After the direct observation of  $F_1$ , the verification of the  $c$ -ring rotation against the  $ab_2$  complex became an important issue. Although around 10 years have passed since the verification of the  $c$ -ring rotation, little progress has been made on the rotary dynamics of  $F_0$ , compared with  $F_1$ , owing to challenges in handling the complicated membrane system and difficulty in stably charging the membrane potential high enough to reverse  $F_1$ . Although detergent-solubilized  $F_0F_1$  was subjected to the rotation assay in ATP hydrolysis conditions in early studies (64, 65), the observed rotation was insensitive to the gold-standard inhibitor of  $F_0$ , dicyclohexyl-carbodiimide (DCCD), implying that the observed rotating is not coupled with the proton translocation of  $F_0$  (64, 66). The subunit interactions of  $F_0$  are weakened in the presence of detergent, which often causes subunit dissociation in biochemical assays (67). Actually, it has been later reported that the rotation in this system is insensitive to mutation at the conserved Arg of the  $a$  subunit (68). Verification of the  $c$ -ring rotation came from biochemical experiments showing that crosslinkage of the  $c$ -ring with the rotor subunits of  $F_1$  ( $\gamma$  and  $\epsilon$  subunits) does not diminish ATP synthesis activity (69), while the  $a$ – $c$  crosslink abolishes ATPase activity coupled with proton translocation (70). Further verification was made by detection of the exchanged cross-link product between the  $a$  and  $c$  subunits, which was probed with a  $^{14}\text{C}$ -labelled  $c$  subunit (71). Single-molecule imaging of rotation under ATP synthesis conditions has also been attempted. The rotation of  $F_0F_1$  reconstituted in liposome was detected from the dipole moment angle of the fluorescent marker dye incorporated into a rotor subunit (72) or Förster

resonance energy transfer (FRET) efficiency between two fluorescent dyes introduced into the stator and rotor subunits (73). A drawback of these experiments is that the membrane potential is transient and, therefore, it is very difficult to correlate the observed rotational velocity with the membrane potential. However, one essential property of  $F_0$  rotation was revealed with the FRET experiment: multiple stepping rotation was detected that was interpreted as  $36^\circ$  steps based on the 10-fold symmetry of the rotor (74). The  $36^\circ$  stepping rotation was later proved in the rotation assay under ATP hydrolysis conditions where a gold nanorod was used as the rotation probe (75).  $F_0F_1$  was reconstituted into a nanodisc of lipid bilayer, and the rotation was monitored from the angle of polarized scattered light along the long axis of the nanorod. However, understanding the dynamics of  $F_0$  rotation is still in its early stages. Experimental systems that allow stable charging of the membrane potential simultaneously with observation of  $F_1$  rotation with high spatiotemporal resolution are highly awaited.

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### Conflict of Interest

None declared.

### References

1. Yoshida, M., Muneyuki, E., and Hisabori, T. (2001) ATP synthase—a marvellous rotary engine of the cell. *Nat. Rev. Mol. Cell. Biol.* **2**, 669–677
2. Kato, Y., Matsui, T., Tanaka, N., Muneyuki, E., Hisabori, T., and Yoshida, M. (1997) Thermophilic F1-ATPase is activated without dissociation of an endogenous inhibitor, epsilon subunit. *J. Biol. Chem.* **272**, 24906–24912
3. Smith, J.B. and Sternweis, P.C. (1977) Purification of membrane attachment and inhibitory subunits of the proton translocating adenosine triphosphatase from *Escherichia coli*. *Biochemistry* **16**, 306–311
4. Sternweis, P.C. and Smith, J.B. (1980) Characterization of the inhibitory (epsilon) subunit of the proton-translocating adenosine triphosphatase from *Escherichia coli*. *Biochemistry* **19**, 526–531
5. Iino, R., Murakami, T., Iizuka, S., Kato-Yamada, Y., Suzuki, T., and Yoshida, M. (2005) Real-time monitoring of conformational dynamics of the epsilon subunit in F1-ATPase. *J. Biol. Chem.* **280**, 40130–40134
6. Saita, E., Iino, R., Suzuki, T., Feniouk, B.A., Kinoshita, K. Jr., and Yoshida, M. (2010) Activation and stiffness of the inhibited states of F1-ATPase probed by single-molecule manipulation. *J. Biol. Chem.* **285**, 11411–11417
7. Suzuki, T., Murakami, T., Iino, R., Suzuki, J., Ono, S., Shirakihara, Y., and Yoshida, M. (2003) F0F1-ATPase/synthase is geared to the synthesis mode by

- conformational rearrangement of epsilon subunit in response to proton motive force and ADP/ATP balance. *J. Biol. Chem.* **278**, 46840–46846
8. Iino, R., Hasegawa, R., Tabata, K.V., and Noji, H. (2009) Mechanism of inhibition by C-terminal alpha-helices of the epsilon subunit of Escherichia coli FoF1-ATP synthase. *J. Biol. Chem.* **284**, 17457–17464
  9. Feniouk, B.A., Suzuki, T., and Yoshida, M. (2006) The role of subunit epsilon in the catalysis and regulation of FOF1-ATP synthase. *Biochim. Biophys. Acta* **1757**, 326–338
  10. Watt, I.N., Montgomery, M.G., Runswick, M.J., Leslie, A.G., and Walker, J.E. (2010) Bioenergetic cost of making an adenosine triphosphate molecule in animal mitochondria. *Proc. Natl Acad. Sci. USA* **107**, 16823–16827
  11. Stock, D., Leslie, A.G., and Walker, J.E. (1999) Molecular architecture of the rotary motor in ATP synthase. *Science* **286**, 1700–1705
  12. Jiang, W., Hermolin, J., and Fillingame, R.H. (2001) The preferred stoichiometry of c subunits in the rotary motor sector of Escherichia coli ATP synthase is 10. *Proc. Natl Acad. Sci. USA* **98**, 4966–4971
  13. Mitome, N., Suzuki, T., Hayashi, S., and Yoshida, M. (2004) Thermophilic ATP synthase has a decamer c-ring: indication of noninteger 10:3 H<sup>+</sup>/ATP ratio and permissive elastic coupling. *Proc. Natl Acad. Sci. USA* **101**, 12159–12164
  14. Meier, T., Polzer, P., Diederichs, K., Welte, W., and Dimroth, P. (2005) Structure of the rotor ring of F-Type Na<sup>+</sup>-ATPase from *Ilyobacter tartaricus*. *Science* **308**, 659–662
  15. Stahlberg, H., Muller, D.J., Suda, K., Fotiadis, D., Engel, A., Meier, T., Matthey, U., and Dimroth, P. (2001) Bacterial Na<sup>+</sup>-ATP synthase has an undecameric rotor. *EMBO Rep.* **2**, 229–233
  16. Meier, T., Matthey, U., von Ballmoos, C., Vonck, J., Krug von Nidda, T., Kuhlbrandt, W., and Dimroth, P. (2003) Evidence for structural integrity in the undecameric c-rings isolated from sodium ATP synthases. *J. Mol. Biol.* **325**, 389–397
  17. Meier, T., Ferguson, S.A., Cook, G.M., Dimroth, P., and Vonck, J. (2006) Structural investigations of the membrane-embedded rotor ring of the F-ATPase from *Clostridium paradoxum*. *J. Bacteriol.* **188**, 7759–7764
  18. Meier, T., Morgner, N., Matthies, D., Pogoryelov, D., Keis, S., Cook, G.M., Dimroth, P., and Brutschy, B. (2007) A tridecameric c ring of the adenosine triphosphate (ATP) synthase from the thermoalkaliphilic *Bacillus* sp. strain TA2.A1 facilitates ATP synthesis at low electrochemical proton potential. *Mol. Microbiol.* **65**, 1181–1192
  19. Preiss, L., Yildiz, O., Hicks, D.B., Krulwich, T.A., and Meier, T. (2010) A new type of proton coordination in an F(1)F(o)-ATP synthase rotor ring. *PLoS Biol.* **8**, e1000443
  20. Seelert, H., Poetsch, A., Dencher, N.A., Engel, A., Stahlberg, H., and Muller, D.J. (2000) Structural biology. Proton-powered turbine of a plant motor. *Nature* **405**, 418–419
  21. Pogoryelov, D., Yu, J., Meier, T., Vonck, J., Dimroth, P., and Muller, D.J. (2005) The c15 ring of the *Spirulina platensis* F-ATP synthase: F1/F0 symmetry mismatch is not obligatory. *EMBO Rep.* **6**, 1040–1044
  22. Diez, M., Zimmermann, B., Borsch, M., Konig, M., Schweinberger, E., Steigmiller, S., Reuter, R., Felekyan, S., Kudryavtsev, V., Seidel, C.A., and Graber, P. (2004) Proton-powered subunit rotation in single membrane-bound F0F1-ATP synthase. *Nat. Struct. Mol. Biol.* **11**, 135–141
  23. Boyer, P.D. (1997) The ATP synthase—a splendid molecular machine. *Annu Rev Biochem.* **66**, 717–749
  24. Wakabayashi, T., Kubota, M., Yoshida, M., and Kagawa, Y. (1977) Structure of ATPase (coupling factor TF1) from a thermophilic bacterium. *J. Mol. Biol.* **117**, 515–519
  25. Kagawa, Y., Sone, N., Yoshida, M., Hirata, H., and Okamoto, H. (1976) Proton translocating ATPase of a thermophilic bacterium. Morphology, subunits, and chemical composition. *J. Biochem* **80**, 141–151
  26. Yoshida, M., Sone, N., Hirata, H., and Kagawa, Y. (1975) A highly stable adenosine triphosphatase from a thermophilic bacterium. Purification, properties, and reconstitution. *J. Biol. Chem.* **250**, 7910–7916
  27. Abrahams, J.P., Leslie, A.G., Lutter, R., and Walker, J.E. (1994) Structure at 2.8 Å resolution of F1-ATPase from bovine heart mitochondria. *Nature* **370**, 621–628
  28. Masaike, T., Koyama-Horibe, F., Oiwa, K., Yoshida, M., and Nishizaka, T. (2008) Cooperative three-step motions in catalytic subunits of F(1)-ATPase correlate with 80 degrees and 40 degrees substep rotations. *Nat. Struct. Mol. Biol.* **15**, 1326–1333
  29. Duncan, T.M., Bulgin, V.V., Zhou, Y., Hutcheon, M.L., and Cross, R.L. (1995) Rotation of subunits during catalysis by Escherichia coli F1-ATPase. *Proc. Natl Acad. Sci. USA* **92**, 10964–10968
  30. Sabbert, D., Engelbrecht, S., and Junge, W. (1996) Intersubunit rotation in active F-ATPase. *Nature* **381**, 623–625
  31. Noji, H., Yasuda, R., Yoshida, M., and Kinosita, K. Jr. (1997) Direct observation of the rotation of F1-ATPase. *Nature* **386**, 299–302
  32. Hayashi, K., Ueno, H., Iino, R., and Noji, H. (2010) Fluctuation theorem applied to F1-ATPase. *Phys. Rev. Lett.* **104**, 218103
  33. Yasuda, R., Noji, H., Kinosita, K. Jr., and Yoshida, M. (1998) F1-ATPase is a highly efficient molecular motor that rotates with discrete 120 degree steps. *Cell* **93**, 1117–1124
  34. Rondelez, Y., Tresset, G., Nakashima, T., Kato-Yamada, Y., Fujita, H., Takeuchi, S., and Noji, H. (2005) Highly coupled ATP synthesis by F1-ATPase single molecules. *Nature* **433**, 773–777
  35. Yasuda, R., Noji, H., Yoshida, M., Kinosita, K. Jr., and Itoh, H. (2001) Resolution of distinct rotational substeps by submillisecond kinetic analysis of F1-ATPase. *Nature* **410**, 898–904
  36. Hirono-Hara, Y., Noji, H., Nishiura, M., Muneyuki, E., Hara, K.Y., Yasuda, R., Kinosita, K. Jr., and Yoshida, M. (2001) Pause and rotation of F(1)-ATPase during catalysis. *Proc. Natl Acad. Sci. USA* **98**, 13649–13654
  37. Shimabukuro, K., Yasuda, R., Muneyuki, E., Hara, K.Y., Kinosita, K. Jr., and Yoshida, M. (2003) Catalysis and rotation of F1 motor: cleavage of ATP at the catalytic site occurs in 1 ms before 40 degree substep rotation. *Proc. Natl Acad. Sci. USA* **100**, 14731–14736
  38. Adachi, K., Oiwa, K., Nishizaka, T., Furuie, S., Noji, H., Itoh, H., Yoshida, M., and Kinosita, K. Jr. (2007) Coupling of rotation and catalysis in F(1)-ATPase revealed by single-molecule imaging and manipulation. *Cell* **130**, 309–321
  39. Watanabe, R., Iino, R., and Noji, H. (2010) Phosphate release in F1-ATPase catalytic cycle follows ADP release. *Nat. Chem. Biol.* **6**, 814–820



40. Watanabe, R., Iino, R., Shimabukuro, K., Yoshida, M., and Noji, H. (2008) Temperature-sensitive reaction intermediate of F1-ATPase. *EMBO Rep.* **9**, 84–90
41. Enoki, S., Watanabe, R., Iino, R., and Noji, H. (2009) Single-molecule study on the temperature-sensitive reaction of F1-ATPase with a hybrid F1 carrying a single beta(E190D). *J. Biol. Chem.* **284**, 23169–23176
42. Ariga, T. (2008) The concerted nature between three catalytic subunits driving the F1 rotary motor. *Biosystems* **93**, 68–77
43. Ariga, T., Muneyuki, E., and Yoshida, M. (2007) F1-ATPase rotates by an asymmetric, sequential mechanism using all three catalytic subunits. *Nat. Struct. Mol. Biol.* **14**, 841–846
44. Okuno, D., Ikeguchi, M., and Noji, H. (2010) Measurement of the conformational state of F(1)-ATPase by single-molecule rotation. *Methods Enzymol.* **475**, 279–296
45. Sielaff, H., Rennekamp, H., Engelbrecht, S., and Junge, W. (2008) Functional halt positions of rotary FOF1-ATPase correlated with crystal structures. *Biophys. J.* **95**, 4979–4987
46. Itoh, H., Takahashi, A., Adachi, K., Noji, H., Yasuda, R., Yoshida, M., and Kinoshita, K. (2004) Mechanically driven ATP synthesis by F1-ATPase. *Nature* **427**, 465–468
47. Rondelez, Y., Tresset, G., Tabata, K.V., Arata, H., Fujita, H., Takeuchi, S., and Noji, H. (2005) Microfabricated arrays of femtoliter chambers allow single molecule enzymology. *Nat. Biotechnol.* **23**, 361–365
48. Collinson, I.R., van Raaij, M.J., Runswick, M.J., Fearnley, I.M., Skehel, J.M., Orriss, G.L., Miroux, B., and Walker, J.E. (1994) ATP synthase from bovine heart mitochondria. In vitro assembly of a stalk complex in the presence of F1-ATPase and in its absence. *J. Mol. Biol.* **242**, 408–421
49. Girvin, M.E., Rastogi, V.K., Abildgaard, F., Markley, J.L., and Fillingame, R.H. (1998) Solution structure of the transmembrane H<sup>+</sup>-transporting subunit c of the F1F0 ATP synthase. *Biochemistry* **37**, 8817–8824
50. Dunn, S.D., McLachlin, D.T., and Revington, M. (2000) The second stalk of Escherichia coli ATP synthase. *Biochim. Biophys. Acta* **1458**, 356–363
51. Sorgen, P.L., Bubb, M.R., and Cain, B.D. (1999) Lengthening the second stalk of F(1)F(0) ATP synthase in Escherichia coli. *J. Biol. Chem.* **274**, 36261–36266
52. Sorgen, P.L., Caviston, T.L., Perry, R.C., and Cain, B.D. (1998) Deletions in the second stalk of F1F0-ATP synthase in Escherichia coli. *J. Biol. Chem.* **273**, 27873–27878
53. Long, J.C., Wang, S., and Vik, S.B. (1998) Membrane topology of subunit a of the F1F0 ATP synthase as determined by labeling of unique cysteine residues. *J. Biol. Chem.* **273**, 16235–16240
54. Valiyaveetil, F.I. and Fillingame, R.H. (1998) Transmembrane topography of subunit a in the Escherichia coli F1F0 ATP synthase. *J. Biol. Chem.* **273**, 16241–16247
55. Wada, T., Long, J.C., Zhang, D., and Vik, S.B. (1999) A novel labeling approach supports the five-transmembrane model of subunit a of the Escherichia coli ATP synthase. *J. Biol. Chem.* **274**, 17353–17357
56. Rubinstein, J.L., Walker, J.E., and Henderson, R. (2003) Structure of the mitochondrial ATP synthase by electron cryomicroscopy. *EMBO J.* **22**, 6182–6192
57. Takeyasu, K., Omote, H., Nettikadan, S., Tokumasu, F., Iwamoto-Kihara, A., and Futai, M. (1996) Molecular imaging of Escherichia coli F0F1-ATPase in reconstituted membranes using atomic force microscopy. *FEBS Lett.* **392**, 110–113
58. Elston, T., Wang, H., and Oster, G. (1998) Energy transduction in ATP synthase. *Nature* **391**, 510–513
59. Junge, W., Lill, H., and Engelbrecht, S. (1997) ATP synthase: an electrochemical transducer with rotary mechanics. *Trends Biochem. Sci.* **22**, 420–423
60. Vik, S.B. and Antonio, B.J. (1994) A mechanism of proton translocation by F1F0 ATP synthases suggested by double mutants of the a subunit. *J. Biol. Chem.* **269**, 30364–30369
61. Dimroth, P., von Ballmoos, C., and Meier, T. (2006) Catalytic and mechanical cycles in F-ATP synthases. Fourth in the Cycles Review Series. *EMBO Rep.* **7**, 276–282
62. Oster, G. and Wang, H. (2003) Rotary protein motors. *Trends Cell Biol.* **13**, 114–121
63. Mitome, N., Ono, S., Sato, H., Suzuki, T., Sone, N., and Yoshida, M. (2010) Essential arginine residue of the F(o)-a subunit in F(o)F(1)-ATP synthase has a role to prevent the proton shortcut without c-ring rotation in the F(o) proton channel. *Biochem J.* **430**, 171–177
64. Panke, O., Gumbiowski, K., Junge, W., and Engelbrecht, S. (2000) F-ATPase: specific observation of the rotating c subunit oligomer of EF(o)EF(1). *FEBS Lett.* **472**, 34–38
65. Sambongi, Y., Iko, Y., Tanabe, M., Omote, H., Iwamoto-Kihara, A., Ueda, I., Yanagida, T., Wada, Y., and Futai, M. (1999) Mechanical rotation of the c subunit oligomer in ATP synthase (F0F1): direct observation. *Science* **286**, 1722–1724
66. Tanabe, M., Nishio, K., Iko, Y., Sambongi, Y., Iwamoto-Kihara, A., Wada, Y., and Futai, M. (2001) Rotation of a complex of the gamma subunit and c ring of Escherichia coli ATP synthase. The rotor and stator are interchangeable. *J. Biol. Chem.* **276**, 15269–15274
67. Tsunoda, S.P., Aggeler, R., Noji, H., Kinoshita, K. Jr., Yoshida, M., and Capaldi, R.A. (2000) Observations of rotation within the F(o)F(1)-ATP synthase: deciding between rotation of the F(o)c subunit ring and artifact. *FEBS Lett.* **470**, 244–248
68. Hosokawa, H., Nakanishi-Matsui, M., Kashiwagi, S., Fujii-Taira, I., Hayashi, K., Iwamoto-Kihara, A., Wada, Y., and Futai, M. (2005) ATP-dependent rotation of mutant ATP synthases defective in proton transport. *J. Biol. Chem.* **280**, 23797–23801
69. Tsunoda, S.P., Aggeler, R., Yoshida, M., and Capaldi, R.A. (2001) Rotation of the c subunit oligomer in fully functional F1Fo ATP synthase. *Proc. Natl Acad. Sci. USA* **98**, 898–902
70. Suzuki, T., Ueno, H., Mitome, N., Suzuki, J., and Yoshida, M. (2002) F(0) of ATP synthase is a rotary proton channel. Obligatory coupling of proton translocation with rotation of c-subunit ring. *J. Biol. Chem.* **277**, 13281–13285
71. Hutcheon, M.L., Duncan, T.M., Ngai, H., and Cross, R.L. (2001) Energy-driven subunit rotation at the interface between subunit a and the c oligomer in the F(O) sector of Escherichia coli ATP synthase. *Proc. Natl Acad. Sci. USA* **98**, 8519–8524
72. Kaim, G., Prummer, M., Sick, B., Zumofen, G., Renn, A., Wild, U.P., and Dimroth, P. (2002) Coupled rotation

- within single F<sub>0</sub>F<sub>1</sub> enzyme complexes during ATP synthesis or hydrolysis. *FEBS Lett.* **525**, 156–163
73. Borsch, M., Diez, M., Zimmermann, B., Reuter, R., and Graber, P. (2002) Stepwise rotation of the gamma-subunit of F<sub>1</sub>F<sub>0</sub>-ATP synthase observed by intramolecular single-molecule fluorescence resonance energy transfer. *FEBS Lett.* **527**, 147–152
74. Duser, M.G., Zarrabi, N., Cipriano, D.J., Ernst, S., Glick, G.D., Dunn, S.D., and Borsch, M. (2009) 36 degrees step size of proton-driven c-ring rotation in F<sub>0</sub>F<sub>1</sub>-ATP synthase. *EMBO J.* **28**, 2689–2696
75. Ishmukhametov, R., Hornung, T., Spetzler, D., and Frasch, W.D. (2010) Direct observation of stepped proteolipid ring rotation in E. coli F<sub>1</sub>F<sub>0</sub>-ATP synthase. *EMBO J.* **29**, 3911–3923